

EVALUATION OF THE METABOLIC PRODUCTION OF THE ENDOPHYTIC FUNGUS *Phomopsis* sp.Jéssica R. da Rocha<sup>a,\*</sup>, Alexandre C. Crispim<sup>a</sup>, João M. da Silva<sup>b</sup>, Antônio E. G. Santana<sup>b</sup> and Luiz C. Caetano<sup>a</sup><sup>a</sup>Instituto de Química e Biotecnologia, Universidade Federal de Alagoas, 57072-900 Maceió – AL, Brasil<sup>b</sup>Centro de Ciências Agrárias, Universidade Federal de Alagoas, 57100-000 Rio Largo – AL, Brasil

Recebido em 08/10/2019; aceito em 04/12/2019; publicado na web em 03/03/2020

Endophytic microorganisms have been shown to be an important source of bioactive compounds, and much has been studied about their metabolites. For example, compounds with different biological activities have been isolated from *Phomopsis* sp. The metabolic study of this genus may lead to a better understanding of the routes of production of metabolites of interest. The aim of this study was to map the metabolic profile of *Phomopsis* sp., isolated as an endophyte of *Syzygium jambolanum* DC, using culture filtrate. The fungus was cultivated for six weeks in potato dextrose medium under hypoxia and at room temperature. The filtrate was collected weekly, and the analyses were performed using proton nuclear magnetic resonance (<sup>1</sup>H NMR) spectroscopy by applying NOESY1D with suppression of the water signal. The occurrence of reductive routes was verified: alcoholic fermentation throughout cultivation and reductive tricarboxylic acid cycle in the cytosol until the third week of cultivation were observed.

Keywords: citric acid cycle; endophytic; *Phomopsis*; alcoholic fermentation; metabolomic study.

## INTRODUCTION

Endophytic microorganisms are present in the inner tissues of plants and coexist in a symbiotic relationship with their hosts without causing damage.<sup>1</sup> On the other hand, endophytes may also behave as pathogens after some kind of external disturbance that, in some way, negatively affects the host. For example, a study of antioxidant activity was carried out in soybean seeds in which compared to seeds infected with *Phomopsis longicolla* and *Cercospora kikuchii*, healthy seeds had a better antioxidant capacity against DPPH (2,2-diphenyl-1-picrylhydrazyl) and ABTS (2,2'-azino-bis(3-ethylbenzothiazoline-6-sulfonic acid)) radicals as well as a higher amount of isoflavones.<sup>2</sup>

Endophytic communities are composed of fungi and bacteria found in a variety of hosts, and it is estimated that many have not yet been identified. Despite their abundance, the relationship between these organisms and their hosts is not yet fully understood. However, many of these endophytes have been applied in agriculture since evidence suggests that they can protect their hosts against pests and pathogenic microorganisms and produce compounds with therapeutic application, such as Taxol and leucinostatin.<sup>3-7</sup>

In general, the most studied endophytes are those isolated from medicinal plants due to their possible symbiotic relationships. Yadav *et al.* verified antioxidant activity in extracts of an endophytic fungus isolated from *Eugenia jambolana* Lam.<sup>8</sup> Volatile compounds with antibacterial activity were identified in an endophyte associated with *Costus spiralis* (Jacq) Roscoe (Costaceae), a plant traditionally used in the treatment of renal diseases.<sup>9</sup> In some cases, the endophyte produces the same compound that is synthesized by the host plant, which makes the endophyte one of the main sources of bioactive compounds, and it is often more advantageous to manipulate these microorganisms in the laboratory instead of the plants, which require cultivation and have management costs. To obtain 3 g of vincristine, for example, 3 kg of dried leaves of *Catharanthus roseus* G. Don are needed. In addition, there are cases in which the yield obtained via endophytic production is greater than that of the plant.<sup>10,11</sup>

The fungus *Phomopsis* sp. has been reported as an endophyte of a variety of plants, including cocoa (*Theobroma cacao* L.) and others of spontaneous growth and medicinal use, such as *Turnera subulata* L.<sup>12,13</sup> In addition, the biological activities associated with this fungus cover a variety of functionalities, such as antiviral activity against tobacco mosaic virus (TMV), production of mycotoxins with amphiphilic potential against phytopathogenic fungi and antibacterial activity against pathogens and phytopathogens.<sup>13-16</sup>

Some of the bioactive compounds produced by *Phomopsis* sp. isolated from medicinal plants include phomoenamides, with moderate antibacterial activity against *Mycobacterium tuberculosis* H37Ra;<sup>17</sup> benzoquinone, with antimicrobial activity against gram-positive bacteria and cytotoxicity against the cancer strain HCT-116;<sup>18</sup> mycoepoxidine, deacetylmethylepoxidine, phomoxydiene A, phomoxene C and cytosporone E, with antimalarial activity against *Plasmodium falciparum* K1 and cytotoxic activity against Vero, KB, MCF-7 and NCI-H187 cells; cytosporone P, with antimalarial activity;<sup>19</sup> 1,5-dihydroxy-3-hydroxyethyl-6-methoxycarbonylxanthone, with cytotoxic activity against A549 and MCF7 tumour cell lines; 1-hydroxy-3-hydroxyethyl-8-ethoxycarbonylxanthone, with cytotoxic activity against the A549 tumour cell line;<sup>20</sup> and phomoxanthone F, with weak anti-HIV activity.<sup>21</sup>

In addition to these biological activities, *Phomopsis* sp. is associated with the ability to degrade plant and soil contaminants, such as phenanthrene.<sup>22</sup>

Given the diversity of applications of the compounds produced by this endophyte, the investigation of the metabolic routes of these compounds contributes to a better understanding of their production, allowing determination of how much, when and what is produced and then enabling production of these metabolites on a large scale.

Nuclear magnetic resonance (NMR) techniques have proven to be an important ally in this type of study and have been supported by chromatography and mass spectrometry.<sup>23,24</sup> This work aimed to identify, through NMR experiments, the metabolites present in the culture filtrate (CF) of *Phomopsis* sp. and to evaluate their production during six weeks of cultivation.

\*e-mail: jessica.rocha@iqb.ufal.br

## EXPERIMENTAL

### Biological material

The endophytic fungus *Phomopsis* sp. was previously isolated from the stem bark of *Syzygium jambolanum*, a plant located at 9°33'18.2"S, 35°46'40.9"W.<sup>25</sup>

### Preparation of culture media

#### Potato dextrose agar culture medium (PDA)

Potato dextrose (20 g) and agar-agar type I (17 g) were dissolved in 1 L of distilled water. The mixture was autoclaved at 121 °C for 17 minutes and then dispensed into previously sterilized petri dishes.

#### Potato dextrose culture medium (PD)

A methodology similar to the preparation of PDA was followed, excluding agar-agar type I; 250 mL Erlenmeyer flasks containing 100 mL of culture medium (CM) were used and autoclaved at 121 °C for 20 minutes.

### Evaluation of metabolic production in the filtered culture of *Phomopsis* sp.

#### Cultivation

The microorganism was initially cultured on PDA for seven days. After this period, inoculation was performed in 40 Erlenmeyer flasks containing PD media previously prepared. The fungus was cultured for eight weeks, with the vials sealed, in the absence of light and without agitation.

#### Sample collection

Aliquots of 1.5 mL were collected weekly from five flasks containing the culture and then refrigerated. This procedure was performed in a sterile horizontal laminar flow chamber. From the collected flasks, the mycelium was separated by simple filtration and dehydrated in an oven at 50 °C for seven days for later weighing. The net fraction was discarded.

#### Mycelial growth

The mycelial growth rate was determined by the average of the five samples collected separately each week, totalling 30 samples. The growth curve was constructed using the dry weight of the mycelium.

#### Sample preparation

A sodium phosphate buffer solution (0.1 mol L<sup>-1</sup>; pH = 7.4) was prepared, and trimethylsilylpropanoic acid (TSP) (1 mmol L<sup>-1</sup>) was added. In each NMR tube, 350 µL of the CF, 350 µL of the buffer solution and 50 µL of D<sub>2</sub>O were added.

#### NMR analysis

The experiments were performed at 25 °C by applying NOESY1D (pulse sequence of noesygppr1d) to suppress the water signal on a Bruker AvanceUltra Shield 400 spectrometer operating at 9.4 T, observing <sup>1</sup>H at 400.12 MHz. The spectrometer was equipped with a 5 mm BBI probe. The parameters were as follows: 64 scans with an acquisition time of 6.29 s, 64 k time domain points distributed in a spectral width of 13.01 ppm and a recycle delay of 6 s.

#### Spectrum analysis

The spectra were processed with TopSpin® 3.5 software using 64 k time domain points. The databases used to identify the metabolites were HMDB (The Human Metabolome Database) and

Chenomx NMR Suite.

#### Statistical NMR data analysis

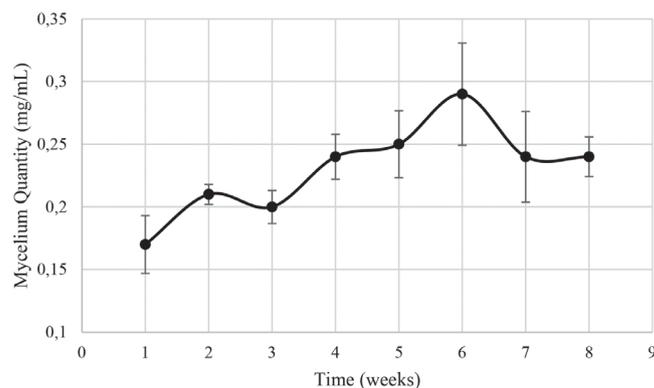
Multivariate statistical analysis was performed using MATLAB® software using a tool developed by Dr. K. Veselkov of Imperial College London. The spectra were first grouped and then aligned and finally normalized.

## RESULTS AND DISCUSSION

### Cultivation and evaluation of mycelial growth

After seven days of culture on PDA, *Phomopsis* sp. presented slight yellowish pigmentation. During the cultivation in PD, the intensification of the yellow colour was observed until the eighth week, in which it presented brownish pigmentation.

According to the mycelial growth curve (Figure 1), the endophyte showed marked growth from the first to the second week and continued to grow until the fourth week. From the fourth week, there was no significant variation in the mycelial mass. In this phase, characterized as a stabilization phase, the accumulation of secondary metabolites may occur.<sup>26</sup>



**Figure 1.** Mycelial growth curve of *Phomopsis* sp. during eight weeks of culture

### Metabolic evolution of the CF

To identify the metabolites of the CF of *Phomopsis* sp., suppression of the water signal by NOESY1D was required.

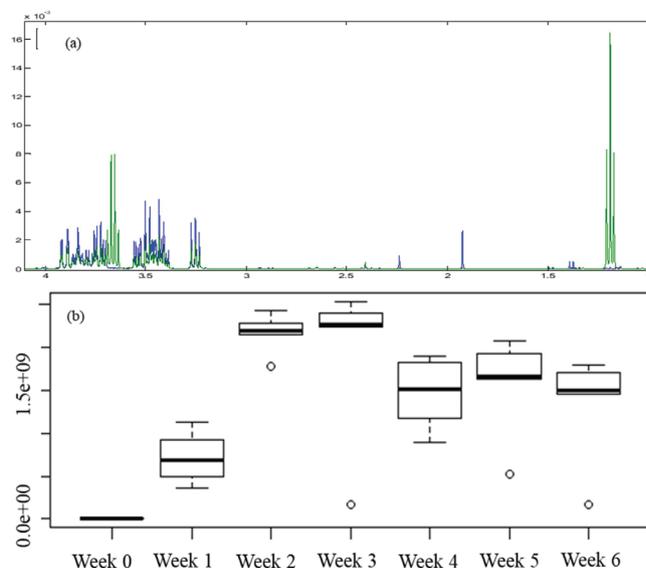
According to the CM spectra, signals attributed to  $\alpha$ - and  $\beta$ -dextrose at 5.24 and 4.65 ppm, respectively, were detected in addition to acetoin (1.37 ppm, *d*, *J* = 7.15 Hz, CH<sub>3</sub>; 2.21 ppm, *s*, CH<sub>3</sub>; 4.42 ppm, *q*, CH), asparagine (2.84 ppm, *m*, CH<sup>+</sup>; 2.94 ppm, *m*, CH<sup>+</sup>; 4.00 ppm, *dd*, *J* = 7.69 and 4.26 Hz, CH), citrate (2.53 ppm, *d*, *J* = 15.88, Ha; 2.66 ppm, *d*, *J* = 15.88, Hb), succinate (2.40 ppm, *s*, CH<sub>2</sub>), pyruvate (2.46 ppm, *s*, CH<sub>3</sub>) (Figure 3), acetate, (1.92 ppm, *s*, CH<sub>3</sub>), ethanol (1.18 ppm, *t*, *J* = 7.00, CH<sub>3</sub>; 3.66 ppm, *q*, *J* = 7.00, CH<sub>2</sub>), oxaloacetate (2.38 ppm, *s*, CH<sub>2</sub>), pyroglutamate (2.02 ppm, *m*, CH<sup>+</sup>; 2.39 ppm, *m*, CH<sub>2</sub>; 2.50 ppm, *m*, CH<sup>+</sup>; 4.17 ppm, *dd*, *J* = 9.02 and 5.83 Hz, CH), tyrosine (6.90 ppm, *dt*, *J* = 2.14, 2.85, 8.56, CH; 7.20 ppm, *dt*, *J* = 2.04, 3.10, 8.56, CH), fumarate (6.53 ppm, *s*, CH) and methanol (3.36 ppm, *s*, CH<sub>3</sub>). The signals detected do not include those of protons of amine and hydroxyl groups present in the structures of these metabolites. This is an expected fact since in the presence of D<sub>2</sub>O, there is spontaneous hydrogen-deuterium exchange.<sup>27</sup>

All chemical shifts and coupling constants were assigned based on the HMDB and Chenomx databases. These assignments were supported by the literature as well.<sup>28-35</sup>

According to the CF spectra from the first to the sixth week of culture, variations in the signal intensities of the metabolites were present. The quantitative variations in these compounds were observed through analysis of variance (ANOVA) performed using a tool in MATLAB® software. A significant increase in the intensity of the ethanol signals (Figure 2a) was observed, indicating the occurrence of fermentation. The ethanol increase was quantified relative to the TSP based on the 1.18 ppm (t) peak area and represented by the boxplot (Figure 2b); these measurements were performed with R software. In addition, the presence of some metabolites that were not detected in CM was observed from the first week of culture (CF1): alanine (1.48 ppm, *d*,  $J = 7.20$  Hz, CH; 3.76 ppm, *q*,  $J = 7.20$  Hz, CH), malate (2.36 ppm, *dd*,  $J = 15.38$  and  $10.12$  Hz, CH<sup>''</sup>; 2.66 ppm, *dd*,  $J = 15.38$  and  $2.90$  Hz, CH<sup>''</sup>); 4.29 ppm, *dd*,  $J = 10.12$  and  $2.80$  Hz, CH), formate (8.46 ppm, *s*, CH) and formaldehyde (9.68 ppm, *s*, CH<sub>2</sub>).

On the other hand, asparagine, detected in CM, was no longer observed in CF1, indicating that the microorganism metabolized it. Figure 3 shows the intensity variations between CM and CF1.

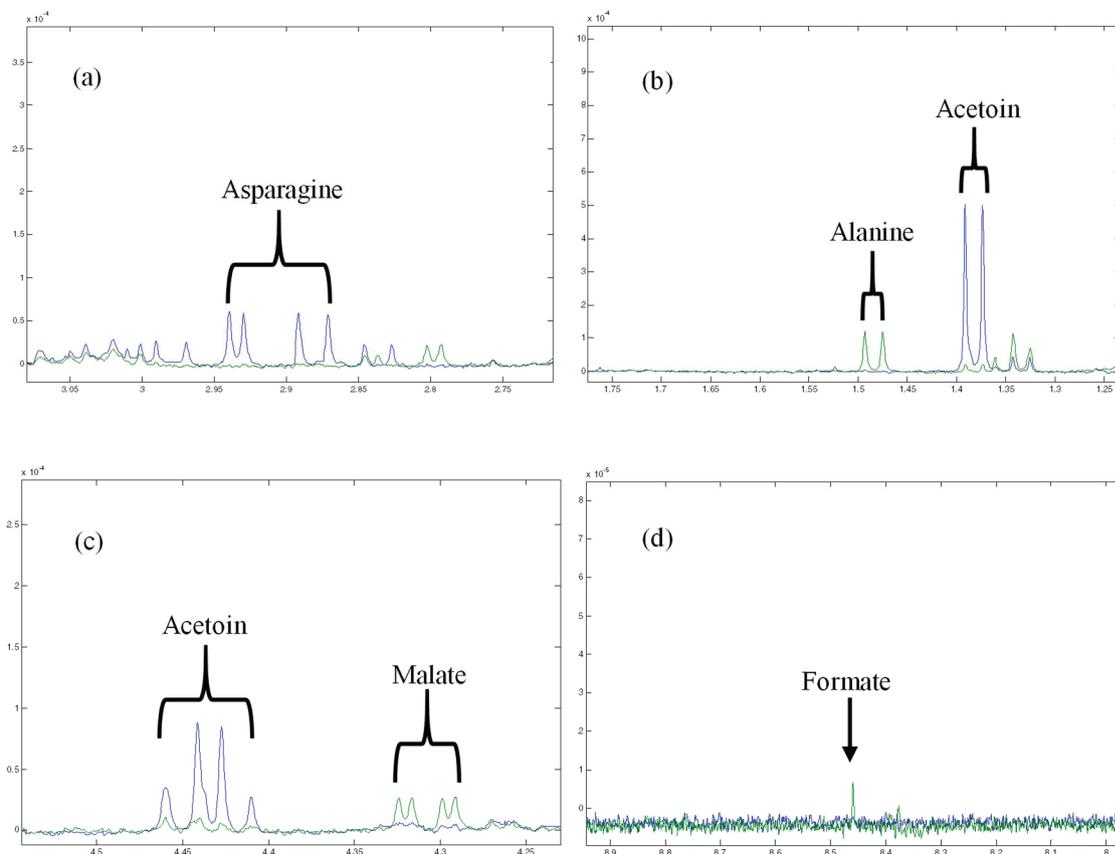
The yellow pigmentation was considerably elevated from the third to the fourth week of cultivation and intensified until the end of the cultivation period. From the fourth week of culture, the quantitative mycelial mass did not present significant development (Figure 1). Considering the spectra of the samples during the six weeks of culture, it was observed that most of the metabolites could not be detected or presented no observable variation from the fourth week. A minority of metabolites could be visibly monitored through the fifth and sixth weeks. Since all the metabolites identified in the CF were primary, it was inferred that from this period (fourth week), secondary metabolism begins using the primary metabolites present for the production of secondary metabolites. Since the sample was very dilute, the intensity of the signals of secondary metabolites



**Figure 2.** Overlapping of the <sup>1</sup>H NMR spectra of CM (in blue) and CF1 (in green) (a); relative quantification of ethanol (b)

whose production was beginning was low, making detection difficult. Table 1 indicates these metabolic variations throughout the culture. The classification of the metabolites arranged in the table was made based on the KEGG PATHWAY database available at <http://www.genome.jp/>.<sup>36</sup>

Variations in the metabolites belonging to the tricarboxylic acid (TCA) cycle were observed through ANOVA performed using a tool in MATLAB® software (supplementary material). The occurrence of the reductive TCA cycle is evident until the third week of culture.<sup>23,37</sup>



**Figure 3.** Overlapping of the <sup>1</sup>H NMR spectra of the CM (in blue) and the culture CF1 (in green): (a) asparagine signals; (b) signs referring to alanine and acetoin; (c) malate and acetoin signals; (d) signal relating to formate

Evans *et al.* first described this route of reductive assimilation of CO<sub>2</sub> by a photosynthetic bacterium.<sup>38</sup> In the present case, non-detection of metabolites such as  $\alpha$ -ketoglutarate and isocitrate also confirms that the reductive route is occurring. From week 0, CM, until the first week, it was possible to observe a decrease in the quantity of oxaloacetate, while the amounts of malate, fumarate and succinate increased. In the following week, in addition to the decrease in the oxaloacetate content, a decrease in the malate and fumarate contents and a consequent increase in the succinate content were observed. From the second to the third week, the oxaloacetate content began to decrease, and the malate content also followed this trend and decreased; however, although the fumarate content had a small increase, it did not present such significant variation. Consequently, there was a decrease in the succinate content, indicating that the metabolites that would lead to its production began to participate more in other routes. From the third to the fourth week, the oxaloacetate content began to rise, but this time, it did not follow the malate content; oxaloacetate was no longer observed from this point, and the fumarate that would provide it was not observed. In this way, the succinate content continued to decay, as this route had been interrupted. Thus, it is concluded that from the interruption of the reductive TCA cycle, from the third to the fourth week, the production routes of secondary metabolites begin to be established. In fact, it is in this period that yellow pigmentation intensifies.

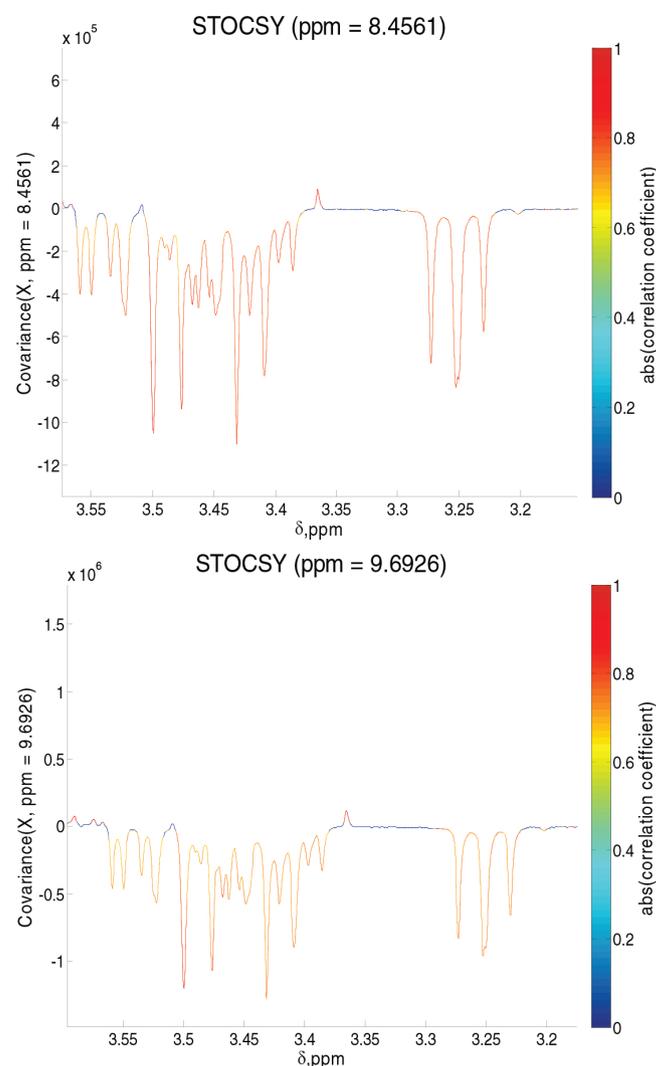
In an anaerobic system, the production of organic compounds can be expected from CO<sub>2</sub>, so this type of system favours reductive routes.<sup>39</sup> In this context, by monitoring the metabolic behaviour of *Phomopsis* sp., it was possible to observe alcoholic fermentation occurrence as well as the evolution of ethanol production up to the sixth week of cultivation (Figure 2).

**Table 1.** Variation in metabolites over 6 weeks of fungal cultivation

Metabolites	Variation per week					
	0-1	1-2	2-3	3-4	4-5	5-6
<i>Metabolism of amino acids and proteins</i>						
Acetoin	↓	↓	↓	*	*	*
Alanine	↑	↓	↓	*	*	*
Asparagine	↓					
Pyroglutamate	~	↑	↑	↓	↓	*
Tyrosine	~	↑	↑	*	*	↑
<i>Carbohydrate metabolism</i>						
Acetate	↓	↑	↑	↑	↑	↑
Ethanol	↑	↑	↑	↓	~	~
Citrate	↑	↑	↓	↓	↓	*
Fumarate	↑	↓	↑	*	*	*
Glucose	↓	↓	↓	↓	↓	↓
Malate	↑	↓	↓	*	*	*
Oxaloacetate	↓	↓	↓	↑	~	~
Succinate	↑	↑	↓	↓	↓	~
<i>Metabolism of methane</i>						
Methanol	↑	↑	↑	~	↓	~
Formaldehyde	↑	↑	~	~	~	↓
Formate	↑	↑	~	*	*	*
<i>Various</i>						
Pyruvate	↓	↑	↑	↓	↓	*

~ = no variation; ↑ = increase; ↓ = decrease; \* = it was not possible to determine presence or variations.

Notably, methanol was monitored until the sixth week of analysis. Its quantity increased until the third week and was maintained until the fourth week, followed by a decrease in the fifth week that was maintained until the last week of cultivation. Both formaldehyde and formate showed increasing quantities until the second week; the latter remained until the third week and could no longer be visualized after that point. On the other hand, the quantity of formaldehyde, which could be monitored until the end, also did not change from the second week, but at the fifth week, it decreased. Statistical total correlation spectroscopy (STOCSY) showed that these three metabolites belong to the same route (Figure 4); the oxidation of methanol leads to formaldehyde and then to the formation of CO<sub>2</sub>.<sup>40</sup> However, there was a limited oxygenation environment (hypoxia), and CO<sub>2</sub> came from the fermentation that occurred throughout the growing period. In fact, there was a tendency for methanol to remain present, while the formate could no longer be observed. The formaldehyde content decreased in the last week, and the methanol content also decreased a week before; however, methanol was present in the last week, indicating that its production was occurring through the reduction of CO<sub>2</sub>.



**Figure 4.** Formate and formaldehyde showed a strong positive correlation and positive covariance with methanol (from the first to the second week)

## CONCLUSION

The identified metabolites of the CF of *Phomopsis* sp. were all from primary metabolism. It is not possible to observe the presence

and/or variation in these metabolites from the fourth week of culture when the yellow pigmentation of the fungus intensifies. It was inferred that during this period, secondary metabolism advances.

This *Phomopsis* sp. NMR metabolomic study performed under the presented conditions showed the occurrence of the reductive TCA cycle and alcoholic fermentation occurring in CF.

## SUPPLEMENTARY MATERIAL

ANOVA images produced using MATLAB® software that verify the variations listed in Table 1 of the metabolites belonging to the TCA cycle are available at <http://quimicanova.s bq.org.br> in PDF file format with free access.

## ACKNOWLEDGMENT

Coordenação de Aperfeiçoamento de Pessoal de Nível Superior (CAPES) for the scholarship provided.

## REFERENCES

- Azevedo, J. L.; Maccheroni Jr., W.; Pereira, J. O.; Araújo, W. L.; *Electron. J. Biotechnol.* **2000**, *3*, 1.
- Lee, J. H.; Hwang, S. R.; Lee, Y. H.; Kim, K.; Cho, K. M.; Lee, Y. B.; *Food Chem.* **2015**, *185*, 205.
- Stierle, A.; Strobel, G.; Stierle, D.; *Science* **1993**, *260*, 5105.
- Strobel, G.; Hess, W. M.; *Chem. Biol. (Oxford, U. K.)* **1997**, *4*, 7.
- Santos, T. T.; Varavallo, M. A.; *UNOPAR Científica Ciências Biológicas e da Saúde* **2011**, *32*, 2.
- Silva, J. M.; Teixeira, R. R. O.; da Rocha, J. R.; dos Santos, T. M. C.; *International Journal of Agriculture, Environment and Bioresarch* **2017**, *2*, 1.
- Yan, J. F.; Broughton, S. J.; Yang, S. L.; Gange, A. C.; *Fungal Ecology* **2015**, *13*.
- Yadav, M.; Yadav, A.; Yadav, J. P.; *Asian Pac. J. Trop. Med.* **2014**, *7*, 1.
- Soares, D. A.; Ascencio, P. G. M.; Leão, G. M. A.; Rodrigues, K. M. T. M.; Pimenta, R. S.; *Journal of Bioenergy and Food Science* **2015**, *2*, 4.
- Bruneton, J.; *Farmacognosia: fitoquímica, plantas medicinales*, 2<sup>nd</sup> ed., Acribia S.A.: Zaragoza, 2001.
- Mussi-Dias, V.; Araújo, A. C. O.; Silveira, S. F.; Rocabado, J. M. A.; Araújo, K. L.; *Braz. J. Med. Plants* **2012**, *14*, 2.
- Rubini, M. R.; Silva-Ribeiro, R. T.; Pomella, A. W. V.; Maki, C. S.; Araújo, W. L.; dos Santos, D. R.; *Int. J. Biol. Sci.* **2005**, *1*, 1.
- Santos, G. B. L.; Caetano, L. C.; Nascimento, A. R. S.; Ramos Sobrinho, R.; Silva, R. M. S., da Silva, J. M.; dos Santos, T. M. C.; *Afr. J. Microbiol. Res.* **2017**, *11*, 17.
- Chapla, V. M.; Biasetto, C. R.; Araujo, A. R.; *Rev. Virtual Quim.* **2013**, *5*, 3.
- Jouda, J.; Tamoku, J.; Mbazona, C. D.; Douala-Meli, C.; Sarkar, P.; Bag, P. K.; Bag, P. K.; Wandji, J.; *BMC Complementary Altern. Med.* **2016**, *16*, 1.
- Tan, Q.; Fang, P.; Ni, J.; Gao, F.; Chen, Q.; *Molecules* **2017**, *22*, 12.
- Rukachaisirikul, V.; Sommart, U.; Phongpaichit, S.; Sakayaroj, J.; Kirtikara, K.; *Phytochemistry* **2008**, *69*, 3.
- Adelin, E.; Servy, C.; Cortial, S.; Lévaïque, H.; Martin, M. T.; Retailleau, P.; Le Goff, G.; Bussaban, B.; Lumyong, S.; Ouazzani, J.; *Phytochemistry* **2011**, *72*, 18.
- Kornsakulkarn, J.; Somyong, W.; Supothina, S.; Boonyuen, N.; Thongpanchang, C.; *Tetrahedron* **2015**, *71*, 48.
- Yang, H. Y.; Gao, Y. H.; Niu, D. Y.; Yang, L. Y.; Gao, X. M.; Du, G.; Hu, Q. F.; *Fitoterapia* **2013**, *91*.
- Hu, H. B.; Hu, H. B.; Luo, Y. F.; Wang, P.; Wang, W. J.; Wu, J. *Fitoterapia* **2018**, *131*.
- Fu, W.; Fu, W.; Xu, M.; Sun, K.; Hu, L.; Cao, W.; Dai, C.; Jia, Y.; *Chemosphere* **2018**, *203*.
- Lindon, J. C.; Nicholson, J. K.; Holmes, E.; *The handbook of metabolomics and metabolomics*, Elsevier: Amsterdam, 2007.
- Baynes, J. W.; Dominiczak, M. H.; *Bioquímica médica*, 4<sup>th</sup> ed., Elsevier: Rio de Janeiro, 2015.
- Lima, S. M. S.; *Monography*, Universidade Federal de Alagoas, Brasil, 2010.
- Nogueira, R. C.; Paiva, R.; Lima, E. C.; Soares, G. A.; Oliveira, L. M.; Santos, B. R.; Emrich, E. B.; Castro, A. H. F.; *Rev. Bras. Plant. Med.* **2008**, *10*, 1.
- Pavia, D. L.; Lampman, G.; Kriz, G.; Vyvyan, J.; *Introdução à espectroscopia*, 2<sup>nd</sup> ed., Cengage Learning: São Paulo, 2015.
- Pomin, V. H. Em *Glycosylation*; Petrescu, S., ed.; InTech: London, 2012, cap. 4.
- Capozzi, F.; Laghi, L.; Belton, P. S.; *Magnetic Resonance in Food Science: Defining Food by Magnetic Resonance*, Royal Society of Chemistry: London, 2015.
- Anet, F. A. L.; Park, J.; *J. Am. Chem. Soc.* **1992**, *114*, 4.
- Somashekar, B. S.; Gowda, G. A. N.; Ramesha, A. R.; Khetrpal, C. L.; *Magn. Reson. Chem.* **2004**, *42*, 636.
- Govindaraju, V.; Young, K.; Maudsley, A. A.; *NMR Biomed.* **2000**, *13*, 129.
- Hinterholzer, A.; Stanojlovic, V.; Cabrele, C.; Schubert, M.; *Anal. Chem.* **2019**, *91*, 14299.
- Corlett, E. K.; Blade, H.; Hughes, L. P.; Sidebottom, P. J.; Walker, D.; Walton, R. I.; Brown, S. P.; *CrystEngComm* **2019**, *21*, 3502.
- Gottlieb, H. E.; Kotlyar, V.; Nudelman, A.; *J. Org. Chem.* **1997**, *62*, 7512.
- <http://www.genome.jp/>, accessed in February 2020.
- Vuoristo, K. S.; Mars, A. E.; Sanders, J. P. M.; Eggink, G.; Weusthuis, R. A.; *Trends Biotechnol.* **2016**, *34*, 3.
- Evans, M. C.; Buchanan, B. B.; Arnon, D. I.; *Proceedings of the national academy of sciences of USA* **1966**, *55*, 4.
- Nelson, D. L.; Cox, M. M.; *Princípios de bioquímica de Lehninger*, 6<sup>th</sup> ed., Artmed: Porto Alegre, 2014.
- Cloarec, O.; Dumas, M. E.; Craig, A.; Barton, R. H.; Trygg, J.; Hudson, J.; Blancher, C.; Gauguier, D.; Lindon, J. C.; Holmes, E.; Nicholson, J.; *Anal. Chem.* **2005**, *77*, 5.