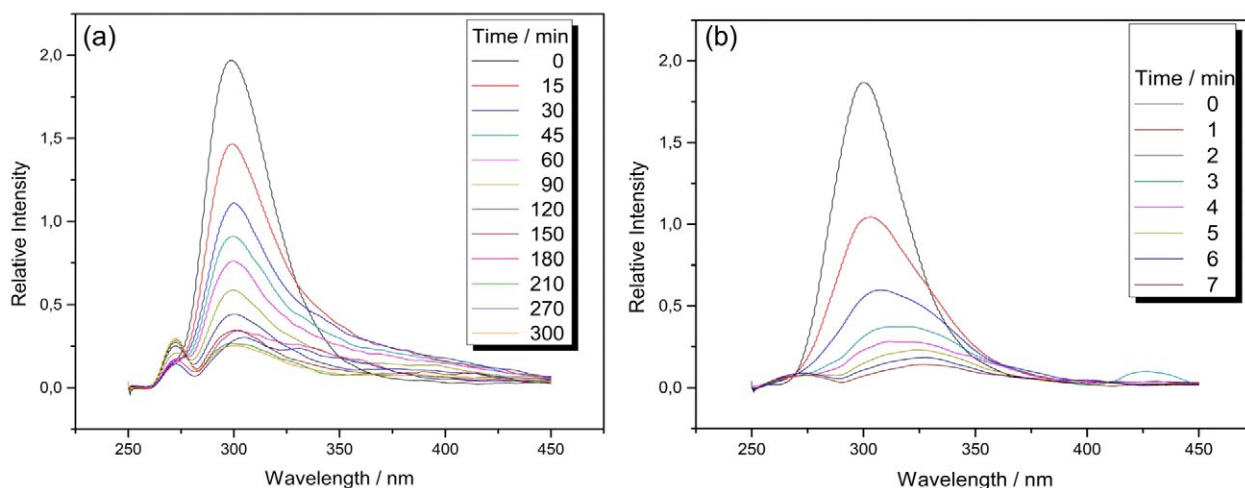


# Supplementary Information

## Application of Headspace Solid-Phase Microextraction and Gas Chromatography-Mass Spectrometry (HS-SPME-GC/MS) on the Evaluation of Degradation Efficiency of Phenolic Compounds and Identification of By-Products from Produced Water

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**Figure S1.** Fluorescence emission spectra depicting the degradation profile for phenol. Initial conditions: (a) UV photolysis: 50.0 mg L<sup>-1</sup> and pH 5.3 (natural pH); (b) 50.0 mg L<sup>-1</sup>; pH 7 and 525 mg H<sub>2</sub>O<sub>2</sub> L<sup>-1</sup> ( $\lambda_{exc.} = 270$  nm;  $\lambda_{emis.} = 300$  nm).

**Table S1.** Monitored response to UV photolysis experiments (data obtained from Statistica 10)

Design: 2 factors, 1 Blocks, 11 Runs (Spreadsheet1.sta)						
time / min	pH	Efficiency removal / % - Means	Efficiency removal / % - Std. Dev.	Efficiency removal / % - N	-95,% - Cnf. Limt	+95,% - Cnf. Limt
1	0	7.0	0.00	0.00000	1	
2	30	3.5	37.28	0.00000	1	
3	30	10.5	14.32	0.00000	1	
4	90	2.0	63.73	0.00000	1	
5	90	7.0	55.80	5.82551	3	41.32721 70.26994
6	90	12.0	29.19	0.00000	1	
7	150	3.5	74.57	0.00000	1	
8	150	10.5	54.03	0.00000	1	
9	180	7.0	76.50	0.00000	1	
All Runs			47.001	24.38	11	30.61640 63.38630

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**Table S2.** Analysis of Variance results to UV photolysis optimization (Data obtained from Statistica 10)

ANOVA; Var.: Efficiency removal / %; R-sqr = ,97322; Adj: ,94644 (Spreadsheet1.sta) 2 factors, 1 Blocks, 11 Runs; MS Residual = 31,85873 DV: Efficiency removal / %					
	SS	df	MS	F	p
(1)Time (min)(L)	4325.157	1	4325.157	135.7605	0.000082
Time (min)(Q)	379.826	1	379.826	11.9222	0.018182
(2)pH (L)	1066.934	1	1066.934	33.4895	0.002169
pH (Q)	100.899	1	100.899	3.1671	0.135260
1L by 2L	1.465	1	1.465	0.0460	0.838688
Error	159.294	5	31.859		
Total SS	5948.375	10			

(1) and (2) factor numbers, (L) linear effect and (Q) Quadratic effect on the response.

**Table S3.** Monitored response to UV/H<sub>2</sub>O<sub>2</sub> experiments (Data obtained from Statistica 10)

Design: 2 factors, 1 Blocks, 11 Runs (Spreadsheet2.sta)							
	H <sub>2</sub> O <sub>2</sub>	pH	% removal Means	% removal Std.Dev.	% removal N	-95,% - Cnf.Limt	+95,% - Cnf.Limt
1	50.0	7.0	98.82	0.000000	1		
2	192.5	3.5	99.13	0.000000	1		
3	192.5	10.5	98.93	0.000000	1		
4	525.0	2.0	99.19	0.000000	1		
5	525.0	7.0	99.36	0.124594	3	99.04780	99.66681
6	525.0	12.0	98.97	0.000000	1		
7	872.5	3.5	99.64	0.000000	1		
8	872.5	10.5	98.51	0.000000	1		
9	1000.0	7.0	99.41	0.000000	1		
All Runs			99.15	0.327408	11	98.93152	99.37144

**Table S4.** Analysis of Variance results to UV/H<sub>2</sub>O<sub>2</sub> optimization (Data obtained from Statistica 10)

ANOVA; Var.: % removal; R-sqr = ,86422; Adj: ,76321 (Spreadsheet2.sta) 2 factors, 1 Blocks, 11 Runs; MS Residual = ,0459094 DV: % removal					
	SS	df	MS	F	p
(1)H <sub>2</sub> O <sub>2</sub> (L)	0.100809	1	0.100809	2.195834	0.198479
H <sub>2</sub> O <sub>2</sub> (Q)	0.107756	1	0.107756	2.347154	0.186081
(2)pH (L)	0.339873	1	0.339873	7.403127	0.041732
pH (Q)	0.127157	1	0.127157	2.769731	0.156946
1L by 2L	0.224422	1	0.224422	4.888373	0.078010
Error	0.162531	5	0.045909		
Total SS	1.372963	10			

(1) and (2) factor numbers, (L) linear effect and (Q) Quadratic effect on the response.

**Table S5.** Mass spectra of the compounds and their retention times

Retention time / min	Compound	Mass Spectra
20.54	2,6-Di-tert-butylbenzoquinone (DBQ)	
20.964	3,5-Di-tert-butylpyrocatechol	
25.201	3,5-Di-tert-butyl-4-hydroxyacetophenone	
25.951	2,5-Di-tert-butylhydroquinone (DTBHQ)	

**Table S5.** Mass spectra of the compounds and their retention times (cont.)

Retention time / min	Compound	Mass Spectra
26.627	3,5-Di-tert-butyl-4-hydroxybenzaldehyde	
24.203	1,4-Naphthoquinone, 6-acetyl-2,5-dihydroxy	
20.721	Benzaldehyde, 2-hydroxy-4-methoxy-3,6-dimethyl	
24.929	2,5-Di-t-amyl-p-benzoquinone (DAQ)	

**Table S5.** Mass spectra of the compounds and their retention times (cont.)

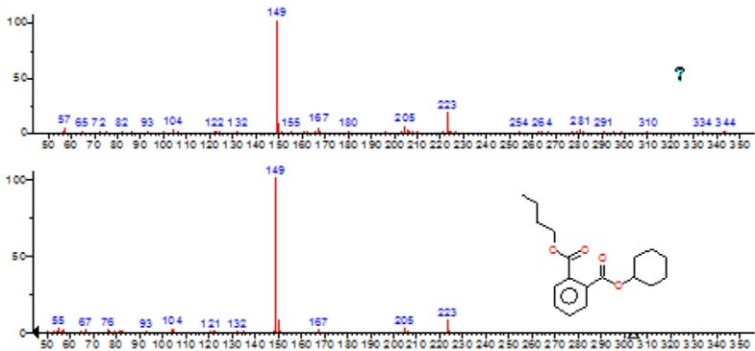
**Table S5.** Mass spectra of the compounds and their retention times (cont.)

Retention time / min	Compound	Mass Spectra
25.749	2'-Methyl-4'-propoxypropiophenone	 <chem>CC(=O)C1=CC=C(C=C1)C=C(C)OCC</chem>
26.939	2-Methyl-1-(2-methyl-4-propoxyphenyl)propan-1-one	 <chem>CC(C)C(=O)C1=CC=C(C=C1)C=C(C)OCC</chem>
27.094	1-(2,6-dimethyl-4-propoxyphenyl)-2-methyl-propan-1-one	 <chem>CC(C)C(=O)C1=CC=C(C=C1)C=C(C)OCC</chem>
23.377	pentanedioic acid, (2,4-di-t-butylphenyl) mono-ester	 <chem>CC(C)(C)C1=CC=C(C=C1)C=C(C)OCCCC(=O)O</chem>

**Table S5.** Mass spectra of the compounds and their retention times (cont.)

Retention time / min	Compound	Mass Spectra
21.620	2,4,6-Tris(1,1-dimethyl)-4-methylcyclohexa-2,5-dien-1-one	 
23.504	Propanoic acid, 2-methyl-, 1-(1,1-dimethyl)-2-methyl-1,3-propanediyl ester	 
24.198	Ethanone, 1,1'-(6-methoxy-2,5-benzofurandiyl)bis	 
28.759	7,9-Di-tert-butyl-1-oxaspiro(4,5)deca-6,9-diene-2,8-dione	 

**Table S5.** Mass spectra of the compounds and their retention times (cont.)

Retention time / min	Compound	Mass Spectra
28.134	1,2-Benzenedicarboxylic acid, butyl cyclohexyl ester	 <p>The mass spectrum shows a base peak at m/z 149. The x-axis ranges from 50 to 350 m/z, and the y-axis represents relative intensity from 0 to 100. The chemical structure of 1,2-Benzenedicarboxylic acid, butyl cyclohexyl ester is shown to the right of the spectrum.</p>