

Nitrofurazone and its Nitroheterocyclic Analogues: a Study of the Electrochemical Behavior in Aqueous Medium

Charles de Lima Brito,^a Gustavo Henrique Goulart Trossini,^a Elizabeth Igne Ferreira^a
and Mauro Aquiles La-Scalea^{*b}^aLaboratório de Planejamento e Síntese de Quimioterápicos Potencialmente Ativos Contra Doenças Negligenciadas (LAPEN), Departamento de Farmácia, Faculdade de Ciências Farmacêuticas, Universidade de São Paulo (USP), Av. Prof. Lineu Prestes, 580, 05508-000 São Paulo-SP, Brazil^bDepartamento de Ciências Exatas e da Terra, Universidade Federal de São Paulo (UNIFESP), Rua Prof. Artur Riedel, 275, 09972-270 Diadema-SP, Brazil

Chemical characterizations of the synthesized analogues

The ¹H NMR spectra were recorded in a Bruker Avance DPX-300 spectrometer, using DMSO-d₆ as solvent and TMS as internal reference. NFS: ¹H NMR (300 MHz, DMSO-d₆) δ 7.63 (d, 1H, *J* 3.66 Hz, H-aromatic), 7.78 (d, 1H, *J* 3.66 Hz, H-aromatic), 7.96 (s, 1H, H-azometine), 8.02 (s, 1H, H-terminal amine), 11.83 (s, 1H, H-amine). NT:

¹H NMR (300 MHz, DMSO-d₆) δ 6.59 (s, 2H, H-amine), 7.43 (d, 1H, *J* 3.72 Hz, H-aromatic), 8.08 (d, 1H, *J* 3.72 Hz, H-aromatic), 8.03 (s, 1H, H-azometine), 10.76 (s, 1H, H-amine). NTS: ¹H NMR (300 MHz, DMSO-d₆) δ 7.53 (d, 1H, *J* 4.23 Hz, H-aromatic), 8.07 (d, 1H, *J* 4.23 Hz, H-aromatic), 8.04 (s, 1H, H-terminal amine), 8.18 (s, 1H, H-terminal amine), 8.45 (s, 1H, H-azometine), 11.82 (s, 1H, H-amine).

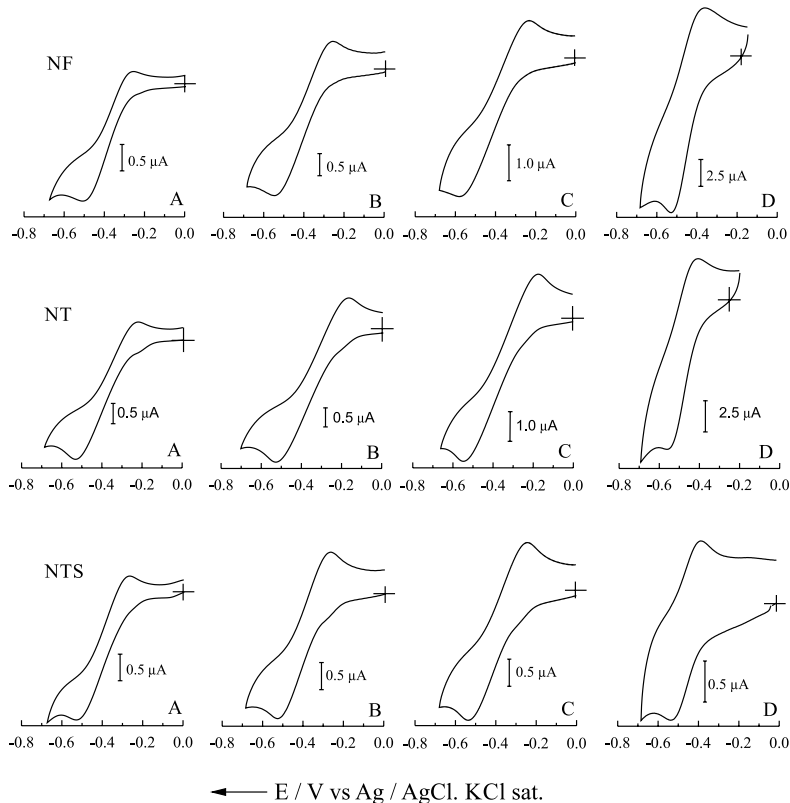


Figure S1. Cyclic voltammograms recorded at pH 10.01. [Compound] = 0.5 mmol L⁻¹. (A) 0.02 V s⁻¹; (B) 0.05 V s⁻¹; (C) 0.1 V s⁻¹; (D) 1.0 V s⁻¹.

*e-mail: mauro.scalea@unifesp.br