

Supplementary Information

Synthesis, Characterization and Catalytic Activity of Two Novel *cis*-Dioxovanadium(V) Complexes: [VO₂(L)] and [VO₂(HLox)]

Natália M. L. Silva,^a Carlos B. Pinheiro,^b Eluzir P. Chacon,^a Jackson A. L. C. Resende,^a
José Walkimar de M. Carneiro,^a Tatiana L. Fernández,^c Marciela Scarpellini^c and
Mauricio Lanznaster^{*,a}

^aInstituto de Química, Universidade Federal Fluminense, Outeiro S. João Batista S/N,
24020-141 Niterói-RJ, Brazil

^bDepartamento de Física, Universidade Federal de Minas Gerais, Av. Antonio Carlos 6627,
Pampulha, 31270-901 Belo Horizonte-MG, Brazil

^cInstituto de Química, Universidade Federal do Rio de Janeiro, Av. Athos da Silveira Ramos 149,
Bl. A, 21941-909 Rio de Janeiro-RJ, Brazil

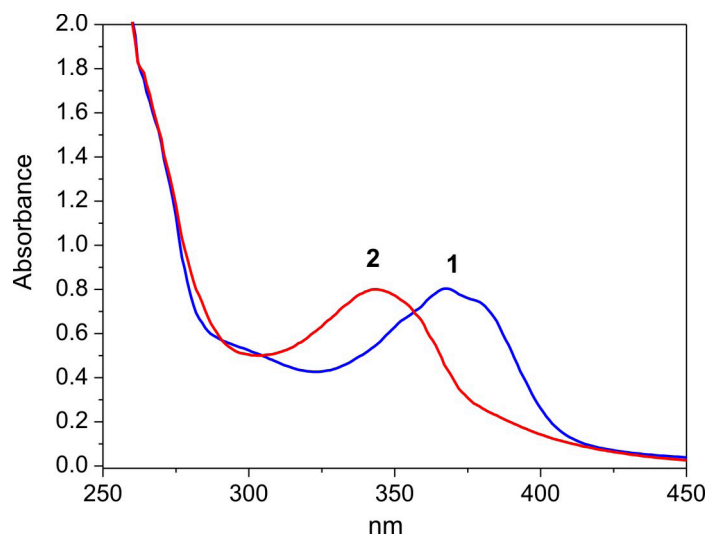


Figure S1. UV-Vis spectra for **1** and **2** recorded in DMSO solutions.

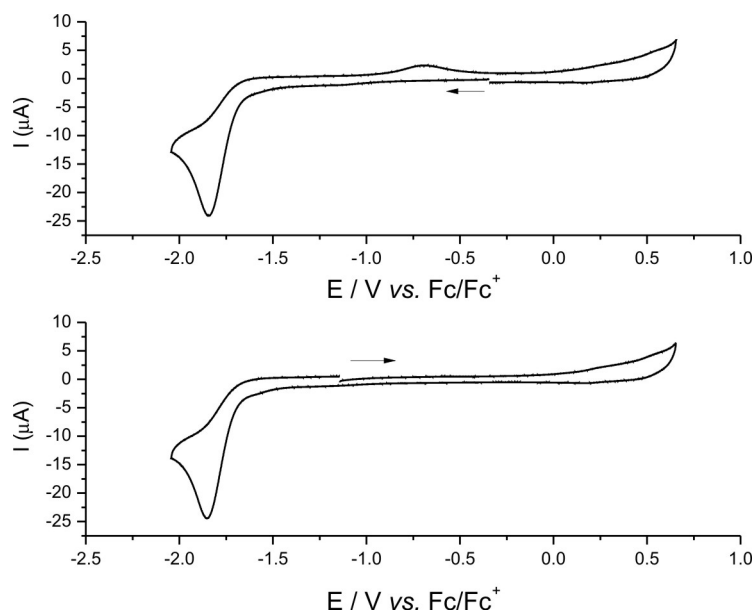


Figure S2. Cyclic voltammograms of complexes **1** in a DMSO/TBAPF₆ 0.1 mol L⁻¹ solution, scan rate = 0.1 V s⁻¹, working electrode = glassy carbon, reference = Ag/AgCl (DMSO/TBAPF₆ 0.1 mol L⁻¹), auxiliary = platinum wire and ferrocene as internal reference. The arrows indicate the scan directions. This figure illustrates the dependence of the process at -0.68 V with the reduction peak at -1.89 V.

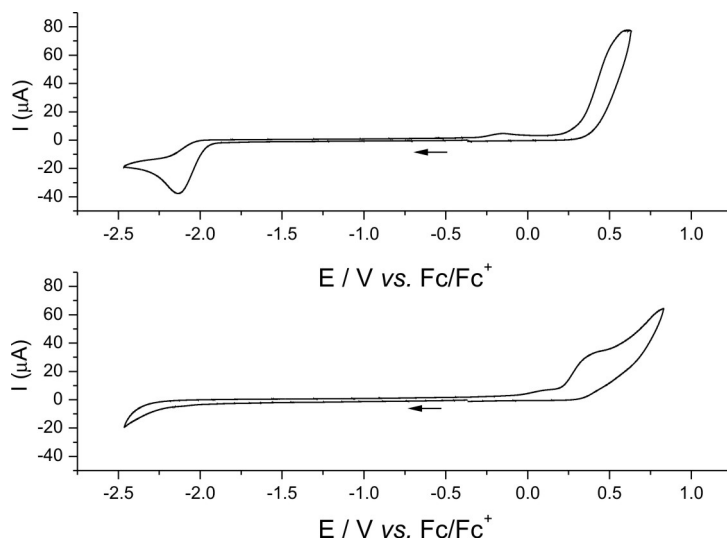


Figure S3. Cyclic voltammograms of the ligands HL (top) and H₂Lox (bottom) in a DMSO/TBAPF₆ 0.1 mol L⁻¹ solution, scan rate = 0.1 V s⁻¹, working electrode = glassy carbon, reference = Ag/AgCl (DMSO/TBAPF₆ 0.1 mol L⁻¹), auxiliary = platinum wire and ferrocene as internal reference. The arrows indicate the scan direction.

Table S1. Summary of the crystal structure data collection and refinement for **1** and **2**

Identification code	[VO ₂ (L)] (1)	[VO ₂ (Lox)] (2)
Empirical formula	(C ₂₁ H ₂₀ N ₃ O ₄ V) ₂ ·H ₂ O	C ₂₁ H ₂₁ N ₄ O ₄ V
Formula weight	876.7	444.36
Temperature	293(2) K	293(2) K
Wavelength	0.71073 Å	0.71073 Å
Crystal system	monoclinic	monoclinic
Space group	P 21/c	P21/c
Unit cell dimensions	$a = 13.316(3)$ Å $b = 14.293(3)$ Å $c = 22.902(7)$ Å $\beta = 113.70(2)^\circ$	$a = 10.126(2)$ Å $b = 15.044(3)$ Å $c = 13.143(3)$ Å $\beta = 97.61(3)^\circ$
Volume	3991.2(17) Å ³	1984.6(7) Å ³
Z	4	4
Density (calculated)	1.459 mg m ⁻³	1.487 mg m ⁻³
Absorption coefficient	0.533 mm ⁻¹	0.537 mm ⁻¹
F(000)	1816	920
Crystal size	0.26 × 0.25 × 0.13 mm ³	0.47 × 0.22 × 0.14 mm ³
θ range for data collection	5.16 to 25.03°	3.61 to 26.00°
Index ranges	-15 ≤ <i>h</i> ≤ 15, -17 ≤ <i>k</i> ≤ 17, -27 ≤ <i>l</i> ≤ 27	-12 ≤ <i>h</i> ≤ 12, -18 ≤ <i>k</i> ≤ 17, -16 ≤ <i>l</i> ≤ 16
Reflections collected	26034	23865
Independent reflections	6986 [R _(int) = 0.0431]	3893 [R _(int) = 0.0351]
Completeness to θ = 26.37°	99.1%	99.6%
Absorption correction	empirical	empirical
Max. and min. transmission	0.948 and 0.897	0.931 and 0.785
Refinement method	Full-matrix least-squares on F ²	Full-matrix least-squares on F ²
Data / restraints / parameters	6986 / 2 / 540	3893 / 0 / 271
Goodness-of-fit on F ²	1.169	1.088
Final R indices [I > 2σ (I)]	R ₁ = 0.0578, wR ₂ = 0.1291	R ₁ = 0.0379, wR ₂ = 0.0966
R indices (all data)	R ₁ = 0.0872, wR ₂ = 0.1386	R ₁ = 0.0502, wR ₂ = 0.1043
Largest diff. peak and hole	0.487 and -0.364 e.Å ⁻³	0.488 and -0.308 e.Å ⁻³