

A Labdane Diterpene from the Aerial Parts of *Egletes viscosa* Less

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Um novo labdano glicosilado, 8 α -hidroxilabd-14(15)-eno-13(S)-O- β -D-ribopiranosídeo, foi isolado das partes aéreas de *Egletes viscosa* Less, juntamente com o 13-*epi*-esclareol, barbatol, tarapacol, espinasterol, ternatina e triacotano. As estruturas dos compostos isolados foram elucidadas pelo uso de métodos espectrométricos e preparação de derivados químicos.

A new labdane glycoside, 8 α -hydroxylabd-14(15)-ene-13(S)-O- β -D-ribopyranoside, was isolated from the aerial parts of *Egletes viscosa* Less, along with 13-*epi*-sclareol, barbatol, tarapacol, spinasterol, ternatin and triacotane. The structures of the isolated compounds were elucidated by the use of spectrometric methods and preparation of chemical derivatives.

Keywords: Asteraceae, *Egletes viscosa*, diterpene glycoside, 8 α -hydroxylabd-14(15)-ene-13(S)-O- β -D-ribopyranoside

Introduction

Egletes viscosa Less (Asteraceae) is an annual herb mostly native to intertropical Americas. It is known in the Northeast of Brazil as “macela” or “macela-da-terra” and grows throughout the margins of river and lakes as soon as the raining season ends. The flower buds are largely sold in the herbal stores and used in the folk medicine as emmenagogue, diaphoretic,¹ stomachic and antidiarrhoeal.²

Previous studies with *E. viscosa* have reported the isolation of biologically active flavonoids and diterpenoids.³⁻¹⁴ Recently, Lee *et al* reported the isolation of labdane diterpene glycosides and flavonoids from the entire plant native from Peru.¹⁵

In continuation of the work with this species, we report now the isolation of the new labdane glycoside 8 α -hydroxylabd-14(15)-ene-13(S)-O- β -D-ribopyranoside **1** in addition to 13-*epi*-sclareol,¹⁶ barbatol,¹⁷ tarapacol,¹⁸ spinasterol,¹⁹ ternatin,^{7,20-23} and triacotane from the aerial

parts of *E. viscosa*. The complete proton and carbon assignments of the known compounds and of the new diterpene **1**, including the unpublished ¹³C and ¹H NMR data of the new derivatives **2-4** were accomplished by the use of a series of 2D NMR experiments such as ¹H-¹H COSY, HMQC and HMBC.

Results and Discussion

Compound **1** was isolated as a white solid. Its molecular formula was determined as C₂₅H₄₄O₆ from the NMR and EI-mass spectroscopy measurements (*m/z* = 440). The absorption band in the IR spectrum at 3400 cm⁻¹ was consistent with hydroxyl groups. Inspection of the ¹H NMR spectrum indicated the presence of three angular methyl groups at δ 0.81 (s, H-19 and H-20) and 0.88 (s, H-18), two methyls attached to oxygenated tertiary carbons at δ 1.11 (s, H-17) and 1.35 (s, H-16), and an anomeric hydrogen at δ 4.32 (d, *J* 6.4 Hz, H-1'). Moreover, the presence of a vinyl moiety was suggested by the signals at δ 5.92 (dd, *J* 17.8, 10.9, H-14); 5.22 (dd, *J* 17.8, 1.2, H-15a) and 5.17 (dd, *J* 10.9, 1.2, H-15b). The ¹³C NMR spectral data of **1** confirmed

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the β -ribopyranosyl unit by the signals at δ 98.6 (C-1'), 73.1 (C-3'), 71.3 (C-2'), 68.2 (C-4') and 65.1 (C-5'),²⁴ together with 20 carbon signals ascribable to the aglycone, that were assigned to five methyls, nine methylenes, seven methine and four non-hydrogenated, two of which oxygen bearing carbons (C-13 and C-8).

The unequivocal assignment of NMR spectral data of compound **1** shown in Table 1 was established by data analysis from COSY, HMQC and HMBC experiments. The spectral data attributed to the aglycone unit showed a very similar feature to those described for sclareol.²⁵ The HMBC experiment confirmed the position of glycosylation at C-13 by the correlations of the anomeric hydrogen at δ 4.32 (H-1') and both methylenes hydrogens at δ 5.22 and 5.17 (H-15a and H-15b) with the oxygenated carbon at δ 80.6 (C-13). The decaline ring system was characterized by long-range correlations of both methyl groups at δ 0.88 (H-18) and 0.81 (H-19) with the carbons at δ 42.0 (C-3) and 56.3 (C-5), as well as the methyl groups at δ 0.81 (H-20) with the carbons at δ 39.9 (C-1), 56.3 (C-5) and 61.7 (C-9). Further,

pertinent correlations of the methyl group at δ 1.11 (H-17) with the carbons at δ 43.7 (C-7), 61.7 (C-9) and 74.0 (C-8) allowed to confirm the allocation of the hydroxyl group at C-8.

A survey in the literature revealed that either (+) or (-) optical rotations can be observed for any kind of sclareol glycosides either from the normal or the enantiomer series, e.g., both the (13R)-labd-14(15)-ene-8,13-diol-13-O- β -fucopyranoside or its aglycone belonging to the normal series show minus signal for their optical rotations,²⁶ but the *ent*-sclareol [*ent*-(13R)-labd-14(15)-ene-8,13-diol] showed (+) signal while its 13-O- β -xylopyranoside derivative showed (-) optical rotation.²⁷ The literature also reports the ¹³C NMR data for sclareol,²⁵ *ent*-sclareol²⁷ and 13-*epi*-sclareol¹⁶. The latter, obtained from *Egletes viscosa* of Peruvian origin, showed $[\alpha]_D^{20} + 4.5^\circ$ (*c* 0.11, CHCl₃).¹⁵

In this work a sclareol entity, **5**, was isolated. Comparison of its optical rotation, + 5.0° (*c* 0.36, CHCl₃), as well as comparison of the ¹³C NMR data with those reported in the literature allowed its identification as 13-

Table 1. NMR assignments for the compounds **1-5** (500/125 MHz)

Carbon	¹ H NMR				¹³ C NMR				
	1	2	3	4	1	2	3	4	5
1	1.68 1.12	1.69 0.96	0.90 1.56	1.57 0.88	39.9	41.3	39.8	39.7	39.9
2	1.62 1.40	1.66 1.29	1.58	1.62 1.29	20.2	21.6	20.5	20.3	20.7
3	1.40 1.18	1.40 1.18	1.13 1.30	1.39 1.13	42.0	43.3	42.4	42.2	42.2
4					39.3	40.5	39.2	39.1	32.7
5	0.94 (dd, <i>J</i> 11.9, 2.1 Hz)	0.96 (dd, <i>J</i> 11.9, 2.1 Hz)	0.92 (m)	0.90 (m)	56.3	57.7	56.3	56.3	56.2
6	1.58 1.30	1.54 1.27	1.21 1.39	1.58 1.41	19.3	20.3	19.1	18.6	19.3
7	1.82 1.40	1.71 1.54	1.34 1.80	1.79 1.42	43.7	43.4	43.3	42.5	44.9
8					74.0	75.4	74.4	74.4	74.4
9	1.30	1.05	1.00	1.03	61.7	63.3	62.0	61.9	62.0
10					32.9	34.3	33.4	33.4	39.4
11	1.65 1.45	1.63 1.44	1.38 1.56	1.47 1.27	18.2	19.6	18.6	18.5	18.6
12	1.82 1.62	1.81 1.44	1.55 1.75	1.47 1.73	44.8	44.8	45.2	43.1	44.3
13					80.6	81.9	80.9	81.0	75.2
14	5.92 (dd, <i>J</i> 17.8, 10.9 Hz)	1.59 (dd, <i>J</i> 7.5, 2.0 Hz)	5.69 (dd, <i>J</i> 17.2, 10.6 Hz)	1.59 1.38	143.2	32.2	142.1	31.4	145.4
15	5.22 (dd, <i>J</i> 17.8, 1.2 Hz) 5.17 (dd, <i>J</i> 10.9, 1.2 Hz)	0.92 (t, <i>J</i> 7.4 Hz)	5.20 (dd, <i>J</i> 17.8, 1.3 Hz) 5.17 (dd, <i>J</i> 10.6, 1.3 Hz)	0.82 (t, <i>J</i> 7.0 Hz)	114.1	8.9	115.3	8.2	112.0
16	1.35 (s)	1.20 (s)	1.30 (s)	1.13 (s)	22.2	24.2	24.4	23.4	24.7
17	1.11 (s)	1.14 (s)	1.14 (s)	1.16 (s)	22.7	24.3	24.5	24.8	24.3
18	0.88 (s)	0.88 (s)	0.85 (s)	0.85 (s)	32.7	34.0	33.5	33.5	33.6
19	0.81 (s)	0.81 (s)	0.77 (s)	0.76 (s)	20.7	22.0	21.6	21.6	21.7
20	0.81 (s)	0.83 (s)	0.75 (s)	0.77 (s)	14.8	16.3	15.7	15.7	15.4
1'	4.32 (d, <i>J</i> 6.4 Hz)	4.42 (d, <i>J</i> 6.4 Hz)	4.56 (d, <i>J</i> 7.3 Hz)	4.62 (d, <i>J</i> 6.5 Hz)	98.6	98.9	96.5	95.9	
2'	3.53	3.52	5.22	5.20	71.3	74.5	69.6	69.8	
3'	3.51	3.51	5.03	5.03	73.1	72.8	70.6	70.7	
4'	3.78	3.79	4.56	5.24	68.2	69.6	67.9	67.8	
5'	3.79	3.98 (dd, <i>J</i> 13.2, 2.5 Hz)	3.60 (d, <i>J</i> 12.5 Hz)		65.1	66.5	63.7	63.7	
	3.49	3.59 (d, <i>J</i> 13.2 Hz)	3.95 (dd, <i>J</i> 12.5, 2.5 Hz)						
1''							169.5	169.5	
2''			2.13 (s)				21.0	21.0	
3''							170.4	170.4	
4''			2.01 (s)				20.8	20.8	
5''							170.6	170.5	
6''			2.05 (s)				21.1	21.1	

epi-sclareol, based mainly on the C-13 chemical shift (δ 75.2), which in the case of sclareol (or *ent*-sclareol) is found at δ 73.0. (Table 1).^{16,25,27} Thus, according to the latter evidences we suggest that the diterpenes here reported belong to the 13-*epi*-sclareol series whose stereochemistry is depicted in Figure 1.

On the basis of the above spectroscopic data, in addition to all optical rotation considerations, compound **1** was identified as the novel labdane diterpenoid 8 α -hydroxyabd-14(15)-ene-13(S)-O- β -D-ribofuranoside. Chemical derivatization by usual methods was used to confirm the proposed structure and yielded the new labdane derivatives **2-4**, to which the ¹H and ¹³C NMR data are presented in Table 1.

The isolated diterpenes 8 α -hydroxyabd-14(15)-en-13(S)-O- β -D-ribofuranoside **1**, barbatol, tarapacol, as well as the derivatives **2-4**, were tested for their cytotoxicity using five tumor cell lines: two human leukemias (HL-60 and CEM), human breast adenocarcinoma (MCF-7), human colon adenocarcinoma (HCT-8) and murine melanoma (B16), but none showed any cytotoxic activity.

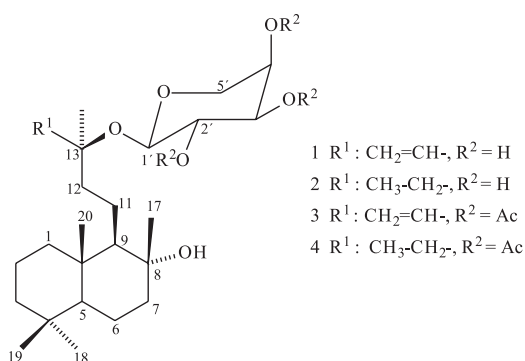


Figure 1. Structures of 8 α -hydroxyabd-14(15)-ene-13(S)-O- β -D-ribofuranoside (**1**), 8 α -hydroxyabd-13(S)-O- β -D-ribofuranoside (**2**), 8 α -hydroxyabd-14(15)-ene-13(S)-O- β -D-2',3',4'-triacetylribofuranoside (**3**) and 8 α -hydroxyabd-13(S)-O- β -D-2',3',4'-triacetylribofuranoside (**4**).

Experimental

General procedures

Melting points were obtained on a Mettler FP82HT apparatus and are uncorrected. IR spectra were recorded using a Perkin Elmer 1000 FT-IR spectrophotometer. Optical rotations were measured on a Perkin Elmer 341 polarimeter. The mass spectra were obtained on a Hewlett-Packard 5971 mass spectrometer by electron impact ionization (70 eV). ¹H and ¹³C NMR spectra were recorded on a Bruker Avance DRX-500 (500 MHz for ¹H and 125 MHz for ¹³C); chemical shifts are given in ppm relative to residual CHCl₃ (7.24 and 77.0 ppm). Silica Gel 60 (Merck, 230-400 mesh) was used for analytical

TLC. Silica gel 60 (Merck, 60 F₂₅₄, 0.2 mm) was used for column chromatography. All compounds were visualized on TLC by spraying with vanillin/perchloric acid/EtOH followed by heating.

Plant material

Aerial parts of cultivated *Egletes viscosa* Less were harvested from garden beds at the Departamento de Fitotecnia of the Universidade Federal do Ceará, Fortaleza, Ceará State. Voucher specimens (36694) have been identified by Dr. Edson Paula Nunes and deposited at the Herbário Prisco Bezerra (EAC), Departamento de Biologia, Universidade Federal do Ceará, Fortaleza, CE, Brazil.

Extraction and isolation

Dried aerial parts of *E. viscosa* (1.35 kg) were pulverized and extracted with hexane at room temperature (3 × 8 L). The solvent was removed under reduced pressure to give the correspondent extract and the mark obtained after hexane extraction was reextracted with EtOH (3 × 8 L).

The hexane extract (12.7 g) was coarsely fractionated on a Silica gel column by elution with hexane, CHCl₃, EtOAc and MeOH. Chromatography of the hexane fraction (1.87 g) using hexane/EtOAc mixtures with increasing polarity yielded triacontane (38.7 mg, 0.3 %). Successive flash chromatography of the CHCl₃ fraction (3.0 g) using 0-100 % hexane/EtOAc mixtures yielded spinasterol (120.0 mg, 0.94 %) and 13-*epi*-sclareol **5** (32.0 mg, 0.25 %). The EtOAc fraction (2.0 g) was further purified over Sephadex LH-20 by elution with MeOH to yield tarapacol (8.9 mg, 0.07 %).

The EtOH extract (15.0 g) was redissolved in a mixture of MeOH:H₂O (1:1 v/v) and submitted to liquid-liquid partition with petrol ether, CHCl₃, EtOAc and MeOH to give four fractions. Chromatography on a Silica gel column of the petrol ether fraction (2.4 g) by elution with hexane, CHCl₃, EtOAc and MeOH yielded the correspondent sub-fractions. Flash chromatography of the CHCl₃ sub-fraction (590.5 mg) using hexane/EtOAc mixtures with increasing polarity, afforded 80 fractions. Preparative chromatography of the fraction 68-73 (43.5 mg) by elution with a mixture of CHCl₃/EtOAc 1:1 yielded compound **1** (19.0 mg). TLC preparative chromatography by elution with hexane/EtOAc (6:4 v/v) of fraction 30-36 (36.0 mg, 0.24 %) yielded barbatol (8.2 mg, 0.05 %). The EtOAc sub-fraction (3.0 g) was further purified over Sephadex LH-20 by elution with MeOH to afford ternatin (6.1 mg, 0.04 %).

Hydrogenation of **1**

Compound **1** (12.0 mg) dissolved in EtOAc (5.0 mL) was added to a suspension containing Pd/Rh (2.0 mg) in MeOH (10.0 mL), that was previously saturated with H₂. The mixture was stirred at room temperature during 30 minutes. Usual work up and SiO₂ column chromatography using hexane/CHCl₃ 2:1 yielded compound **2** (10.0 mg, 83.3 %) as a white solid.

Acetylation of **1**

To a solution of compound **1** (5.0 mg) in pyridine (1.0 mL) were added Ac₂O (1.0 mL) and a catalytic amount of DMAP. The mixture was stirred for 30 minutes at room temperature. Subsequent workup afforded a residue that was chromatographed using hexane/CHCl₃ (3:1) as eluent to yield compound **3** (6.0 mg, 94.0 %) as a yellowish solid.

Hydrogenation of **2**

Compound **2** (10.0 mg) dissolved in EtOAc (4.0 mL) was added to a suspension containing Pd/Rh (2.0 mg) in MeOH (80.0 mL) that was previously saturated with H₂. The mixture was stirred at room temperature during 30 minutes. Usual work up and SiO₂ column chromatography using hexane/CHCl₃ 2:1 yielded compound **4** (8.0 mg, 80.0 %) as a white solid.

8 α -hydroxylabd-14(15)-ene-13(S)-O- β -D-ribofuranoside (**1**)

White solid; m.p. 89.7 – 91.0 °C; [α]_D²⁰: +10.0° (MeOH; *c* 0.02); IR (film) ν_{\max} /cm⁻¹: 3400, 2927, 2867, 1647, 1459, 1388, 1131, 1073, 1002, 918, 784; EIMS *m/z* (rel. int.): 440 [M⁺](3), 290 (8), 273 (15), 257 (6), 245 (3), 232 (2), 217 (8), 203 (13), 191 (59), 177 (75), 163 (25), 149 (14), 137 (14), 123 (20), 109 (39), 95 (58), 81 (79), 69 (44), 43 (100), 41 (71).

8 α -hydroxylabd-13(S)-O- β -D-ribofuranoside (**2**)

White solid; m.p. 71.7 – 74.2 °C; [α]_D²⁰: +7.0° (MeOH; *c* 0.28); IR (film) ν_{\max} /cm⁻¹: 3400, 2927, 2867, 1647, 1459, 1388, 1131, 1073, 1002, 918, 784; EIMS *m/z* (rel. int.): 442 [M⁺](4), 292 (27), 275 (100), 259 (17), 245 (41), 234 (17), 221 (14), 205 (21), 191 (43), 177 (57), 163 (25), 149 (30), 137 (36), 123 (38), 109 (78), 95 (68), 95 (69), 69 (60), 43 (74), 41 (37).

8 α -hydroxylabd-14(15)-ene-13(S)-O- β -D-2',3',4'-triacetylribofuranoside (**3**)

Colorless resin; [α]_D²⁰: +15.0° (CHCl₃; *c* 0.02); IR (film) ν_{\max} /cm⁻¹: 3366, 2925, 2852, 1748, 1493, 1380, 1347, 1131, 1092, 1073, 1002, 944, 823, 702; EIMS *m/z* (rel. int.): 566 [M⁺](5), 292 (14), 275 (63), 259 (11), 245 (31), 234 (3), 221 (7), 204 (27), 191 (37),

177 (56), 163 (27), 149 (31), 137 (32), 123 (28), 109 (42), 95 (58), 73 (83), 69 (71), 43 (100), 41 (52).

8 α -hydroxylabd-13(S)-O- β -D-2',3',4'-triacetylribofuranoside (**4**)

Colorless resin; [α]_D²⁰: +13.0° (CHCl₃; *c* 0.02); IR (film) ν_{\max} /cm⁻¹: 3322, 2930, 2823, 1752, 1333, 1147, 1096, 1081, 952, 735; EIMS *m/z* (rel. int.): 568 [M⁺](3), 292 (25), 275 (79), 259 (100), 245 (25), 234 (4), 217 (15), 199 (25), 191 (31), 177 (38), 157 (39), 151 (13), 139 (82), 123 (24), 109 (51), 97 (63), 81 (31), 69 (32), 43 (100), 41 (7).

Determination of Cytotoxicity

All compounds (0.39 to 25 μ g/mL) were tested for cytotoxic activity at the *Departamento de Fisiologia e Farmacologia* of the *Universidade Federal do Ceará*. Five tumor cell lines were used (National Cancer Institute, Bethesda, MD, USA): B=16 (murine melanoma), HCT-8 (human colon), MCF-7 (human breast), CEM and HL-60 (leukemia) after 72 hours of incubation. Doxorubicin (0.01 to 0.58 μ g/mL) was used as a positive control. The general viability of cultured cells was determined by reduction of the yellow dye 3-(4,5-dimethyl-2-thiazolyl)-2,5-diphenyl-2H-tetrazolium bromide (MTT) to a blue formazan product as described by Mosmann.²⁸

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Supplementary Information

Supplementary data including spectral data of all compounds are available free of charge at <http://jbc.sbq.org.br>, as PDF file.

References

1. Braga, R.; *Plantas do Nordeste, especialmente do Ceará*, Imprensa Oficial: Fortaleza, Brazil, 1960.
2. Correia, P.; *Dicionário de Plantas Úteis do Brasil e das Exóticas Cultivadas*, Instituto Brasileiro de Desenvolvimento Florestal: Rio de Janeiro, Brazil, 1984.
3. Souza, M. F.; Rao, V. S.; Silveira, E. R.; *Braz. J. Med. Biol. Res.* **1992**, *25*, 1029.
4. Souza, M. F.; Cunha, G. M. A.; Fontenele, J. B.; Viana, G. S. B.; Rao, V. S.; *Phytother. Res.* **1994**, *8*, 478.
5. Melo, C.L.; Maia, S. B.; Souza, M. F.; Rao, V. S.; Lima, M. A. S.; Silveira, E. R.; *Fitoterapia* **1993**, *64*, 306.

6. Rao, V. S.; Figueiredo, E. G.; Melo, C. L.; Viana, G. S. B.; Menezes, D. B.; Matos, M. S. F.; Silveira, E. R. *Pharmacology* **1994**, *48*, 392.
7. Lima, M. A. S.; Silveira, E. R.; Marques, M. S. L.; Santos, R. H. A.; Gambardela, M. T. P.; *Phytochemistry* **1996**, *41*, 217.
8. Rao, V. S. N.; Paiva, L. A. F.; Souza, M. F.; Campos, A. R.; Ribeiro, R. A.; Brito, G. A. C. Souza, M. F.; Rao, V. S. N.; Silveira, E. R.; *Phytomedicine* **1997**, *4*, 27.
9. Rao, V. S. N.; Santos, F. A.; Sobreira, T. T.; Souza, M. F.; Melo, C. L.; Silveira, E. R.; *Planta Med.* **1997**, *63*, 146.
10. Souza, M. F.; Rao, V. S. N.; Silveira, E. R.; *Phytother. Res.* **1998**, *12*, 28.
11. Guedes, M. M.; Cunha, A. N.; Silveira, E. R.; Rao, V. S. N.; *Planta Med.* **2002**, *68*, 1044.
12. Teixeira, M. J.; Silveira, E. R.; *Planta Med.* **2003**, *69*, 851.
13. Vieira, G. A. B.; Lima, M. A. S.; Bezerra, A. M.; Silveira, E. R.; *J. Braz. Chem. Soc.* **2006**, *17*, 43.
14. Melo, C. M.; Maia, J. L.; Cavalcante, I. J. M.; Lima, M. A. S.; Vieira, G. A. B.; Silveira, E. R.; Rao, V. S. N.; Santos, F. A.; *Planta Med.* **2006**, *72*, 584.
15. Lee, D.; Li, C.; Graf, T. N.; Vigo, J. S.; Graham, J. G.; Cabieses, F.; Farnsworth, N. R.; Cordell, G. A.; Kinghorn, A. D.; Kroll, D. J.; Wani, M. W.; Oberlies, N. H.; *Planta Med.* **2005**, *71*, 792.
16. Stierle, D. B.; Stierle, A. A.; Larsen, R. D.; *Phytochemistry* **1988**, *27*, 517.
17. Rivero-Cruz, I.; Trejo, J. L.; Aguilar, M. I.; Bye, R.; Mata, R.; *Planta Med.* **2000**, *66*, 734.
18. Zhou, L.; Fuentes, R. R.; Hoffmann, J. J.; Timmermann, B. N.; *Phytochemistry* **1995**, *40*, 1201.
19. Henry, M.; Chatalat-Dublanche, I.; *Planta Med.* **1995**, *4*, 322.
20. Vieira, M. M.; Macedo, F. Y. B.; Belarmino Filho, J. N.; Costa, A. C. L. V.; Cunha, A. N.; Silveira, E. R.; Brito, G. A. C.; Ribeiro, R. A.; *Phytother. Res.* **2004**, *18*, 135.
21. Pessoa, C.; Silveira, E. R.; Lemos, T. L. G.; Wetmore, L. A.; Moraes, M. O.; Leyva, A.; *Phytother. Res.* **2000**, *14*, 187.
22. Souza, M. F.; Rao, V. S. N.; Silveira, E. R.; *Phytother. Res.* **1998**, *12*, 557.
23. Souza, M. F.; Tome, A. R.; Rao, V. S. N.; *Pharm. Pharmacol.* **1999**, *51*, 125.
24. Breitmaier, E.; Voelter, N.; *¹³C NMR Spectroscopy Methods and Applications*. Verlag Chemie: Düsseldorf, Germany, 1974.
25. Kouzi, S.A.; McChesney, J. D.; *J. Nat. Prod.* **1991**, *54*, 483.
26. Lee, S.O.; Choi, S. Z.; Choi, S. U.; Lee, K. C.; Chin, Y. W.; Kim, J.; Kim, Y. C.; Lee, K. R.; *J. Nat. Prod.* **2005**, *68*, 1471.
27. Torrenegra, R.; Robles, J.; Waibel, R.; Lowel, M.; Achenbach, H.; *Phytochemistry* **1994**, *35*, 195.
28. Mosmann, T.; *J. Immunol. Methods* **1983**, *16*, 55.

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