

Metabolomics-Based Discovery of Biomarkers with Cytotoxic Potential in Extracts of *Myracrodruon urundeuva*

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Myracrodruon urundeuva (“aroeira-do-sertão”) is a species threatened with extinction due to anthropogenic exploitation. Phytochemical analysis of bark, branch and leaf extracts revealed the presence of several compounds such as flavonoids, phenols, tannins, quercetin derivatives and anacardic acids. Dereplication methodology was performed to tentatively identify 50 compounds analyzed by ultra-performance liquid chromatography coupled with an electrospray ionization quadrupole time-of-flight mass spectrometry operating in MS^E mode (UPLC-QTOF-MS^E). The extracts exhibited anti-tumor effect in cancer cells HCT-116 (colorectal), SF-295 (glioblastoma), HL-60 (leukemia), and RAJI (leukemia). Also, these results correlate with the principal component analysis (PCA) data that identified three distinct groups indicating, efficiently, metabolic differences between organs of *M. urundeuva*. Through discriminatory analysis of the orthogonal partial least squares (OPLS-DA), the variable of importance in the projection (VIP) and S-Plot, we were able to determine 30 potential biomarkers. The fingerprint of hydroethanolic extracts was correlated with the cytotoxicity assay and demonstrated a significant difference in the composition of plant extract.

Keywords: *Myracrodruon urundeuva*, dereplication, UPLC-MS/MS, chemometrics, cytotoxic activity

Introduction

Myracrodruon urundeuva Fr.All. (Anacardiaceae family), popularly known as “aroeira-do-sertão”, is a medicinal tree found in several regions of Brazil, especially in the caatinga.¹ Currently, it is included in the official list of Brazilian flora species threatened with extinction² in the vulnerable category due to indiscriminate use of the species for several purposes in the wood and pharmaceutical area.³

The plant raises the researchers interest due to anti-inflammatory properties of its extracts, notably associated with the presence of bioactive phenolic compounds such as tannins, polyphenols, ellagitannins and, mainly, dimeric chalcones.⁴ Previous studies have shown that the chemical properties of these substances may be associated with antitumor activity in lung cells and leukemia.⁵ Pharmacological studies revealed a wide variety of pharmacological activities including cytotoxic,^{6,7} anti-inflammatory and analgesic.⁸ Besides, *M. urundeuva* may prevent cancer indirectly due to antioxidant and anti-inflammatory activity of its compounds.^{4,8-10}

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Previous studies have reported antitumor activity of ethanolic extracts from different sections of the plant. Extract dilutions yielded a half maximal inhibitory concentration (IC_{50}) between 9.5-16.7 $\mu\text{g mL}^{-1}$ against leukemic HL-60 and, among other types such as SF-295 glioblastoma of IC_{50} 17.3-36.3 $\mu\text{g mL}^{-1}$. The activity of this extract was reported to occur via an apoptotic mechanism, which results in a reduction of cell numbers, cell volume, and viability in addition to internucleosomal DNA fragmentation.⁷

Pessoa *et al.*⁵ reported the action of ethanolic extracts of the *Myracrodruon urundeuva* leaf against the HL-60 and SW-1573 lines with IC_{50} of 7.4 and 8.5 $\mu\text{g mL}^{-1}$, respectively. Therefore, the literature shows several shreds of evidence of significant antitumor activity from *M. urundeuva* extracts, which motivated the study of its chemical composition. No previous research has been conducted with the identification of compounds associated with the biological activity of *M. urundeuva* extracts. The metabolomic study of plant samples is of great importance when one wants trying to associate certain bioactivity with the chemical composition of the extract. In this regard, the metabolomics focuses on the study of low molecular weight compounds that may be established as biomarkers¹¹ by means of metabolic fingerprinting and profiling.¹² This area of study covers a set of analyses from extraction methods to the statistical analysis of the data in order to identify molecules that can function as biomarkers or from genome alteration.¹³ This is because it can be used in different spheres, such as metabolic fingerprint, metabolic profile, and metabolomics.¹² The ultra-performance liquid chromatography (UPLC) coupled to high-resolution mass spectrometry (HRMS) has the ability and sensitivity to provide a high-resolution mass spectrum for a complex matrix as a plant extract. Therefore, it is widely used in metabolomics studies involving identification of substances from high complexity extracts.¹⁴⁻¹⁶

Ultra-performance liquid chromatography coupled with an electrospray ionization quadrupole time-of-flight mass spectrometry operating in MS^E mode (UPLC-QTOF- MS^E) allied to chemometric analysis is very useful for the identification of compounds by comparison of different matrices. Further analyses as principal component analysis (PCA) and orthogonal projections to latent structures discriminant analysis (OPLS-DA) identify groups that differ from each other, as well as presenting the responsible components that cause these differences, which are recognized as discriminant.¹⁷ Thus, these analyses help to provide information about compounds that could be used as diagnostic to each sample type, therefore potential biomarkers.

Present work aimed to explore differences in metabolic fingerprints of *M. urundeuva* leaves, branches, and bark ethanolic extracts by using UPLC-QTOF- MS^E and multivariate modeling (PCA and OPLS-DA) in order to search associations between chemical composition and cytotoxic effect.

Experimental

Plant material

Samples of leaf, bark and branch from *M. urundeuva* were collected from naturally occurring young plants from the Embrapa semi-arid experimental field, close to the border between the municipalities of Petrolina and Lagoa Grande (Pernambuco State, Brazil, 09°04'16.4"S, 40°19'5.37"W) on August 24, 2016, between 9 and 10 o'clock in the morning. The voucher specimens have been deposited in the Herbarium with number HTSA4978. Samples of leaves, bark, and branches were collected in biological quintuplets (in five different trees) taking into account the four quadrants of the tree, in the north, south, east and west directions. The four quadrants were assembled in single samples for each section of the plant. At the time of collection, the liquid nitrogen cooling process was performed at $-80\text{ }^{\circ}\text{C}$. After that, the material was dried in a forced circulation oven at $40\text{ }^{\circ}\text{C}$ for 168 h (one week). Prior to extraction, the samples were ground in a knife mill and stored in a plastic bag at room temperature.

Chemicals

The solvents used were from LiChrosolv[®] of the Sigma-Aldrich Chemical Company (St. Louis, MO, USA). In all methods, high purity Milli-Q water (Billerica, MA, USA) was used. The standards for chlorogenic acid and corilagin were obtained from Sigma-Aldrich Chemical Company (St. Louis, MO, USA) and urundeuve A and B were previously isolated by our laboratory.

Sample preparation

The method used was adapted for the preparation of extracts by liquid-liquid partition.^{18,19} Leaves, branches, and bark (50 mg) were added in Falcon (15 mL) tube and extracted with 4 mL hexane, at room temperature, for 20 min in ultrasound batch. Afterward, 4 mL of EtOH:H₂O (7:3) solution was added. The samples were extracted again with hexane, and the hydroethanolic partition was collected to yield the corresponding EtOH extract. Finally, a 1 mL aliquot of the lower (hydroethanolic) phase was filtered

(0.20 μm polytetrafluoroethylene (PTFE)), collected in flasks and stored at $-80\text{ }^{\circ}\text{C}$ until further UPLC analysis.

Chromatographic conditions

The analysis was performed using an Acquity UPLC (Waters) system, coupled with a quadrupole/TOF (Waters) system. A Waters Acquity UPLC BEH column ($150 \times 2.1\text{ mm}$, $1.7\text{ }\mu\text{m}$) was used, with the column temperature set at $40\text{ }^{\circ}\text{C}$. The binary gradient elution system consisted of 0.1% formic acid in water (A) and 0.1% formic acid in acetonitrile (B). The UPLC elution conditions were optimized as follows: linear gradient from 2 to 95% B (0-15 min), 100% B (15-17 min), 2% B (17.01 min), 2% (17.02-19.01 min), a flow of 0.4 mL min^{-1} , and a sample injection volume of $5\text{ }\mu\text{L}$.

Mass spectrometry conditions

The chemical profiling of *M. urundeuva* leaves, branches, and bark extracts was performed by coupling the Waters Acquity UPLC system to the QTOF mass spectrometer (Waters, Milford, MA, USA) with the electrospray ionization interface (ESI) in positive and negative ionization modes. The ESI⁺ and ESI⁻ data was acquired in the range of 110-1180 Da, with a fixed source temperature of $120\text{ }^{\circ}\text{C}$, and a desolvation temperature of $350\text{ }^{\circ}\text{C}$. A desolvation gas flow of 350 L h^{-1} was used for the ESI⁺ mode and the 500 L h^{-1} for the ESI⁻ mode. The capillary voltage was 3 kV. Leucine enkephalin was used as a lock mass. The MS model used was Xevo G2-XS QTOF. The spectrometer operated with MS^E centroid programming using a tension ramp from 20 to 40 eV. The instrument was controlled by MassLynx 4.1 software (Waters Corporation).

Chemometric data analysis

The UPLC-MS data of all samples were analyzed using the MarkerLynx XS software²⁰ to identify potential discriminatory chemical markers in different extracts. For data collection, the method parameters were set as retention time (t_{R}) range, 0.88-17.0 min, and mass range of 110-1180 Da. For data analysis, a list composed of the identities of the detected peaks was generated using retention time (t_{R})-mass data (m/z) pairs as the identifier for each peak. An arbitrary ID was assigned to each of this $t_{\text{R}}-m/z$ pairs based on their order of elution from the UPLC system. The ion intensity for each detected peak was normalized against the sum of the peak intensities within that sample. Ion identification was based on the t_{R} and m/z

values. The resulting three-dimensional data comprising peak number ($t_{\text{R}}-m/z$ pair), sample name, and ion intensity were analyzed by PCA and OPLS-DA using MarkerLynx.²⁰

Cytotoxicity of leaf, bark, and branch samples from *M. urundeuva*

Cell lines and cultures

Cytotoxicity tests were performed against HCT-116 and SW-620 (colorectal), SF-295 (glioblastoma), HL-60 and RAJI (leukemia), PC3 (prostate) and L929 (murine fibroblast) cell lines, which were obtained from the National Cancer Institute (Washington, DC, USA). All cells were cultured in Roswell Park Memorial Institute (RPMI) 1640, except for L929, which was cultivated in Dulbecco's Modified Eagle Medium (DMEM). Both mediums were supplemented with 10% fetal bovine serum (FBS) and 1% antibiotics (100 U mL^{-1} penicillin and $100\text{ }\mu\text{g mL}^{-1}$ streptomycin) at $37\text{ }^{\circ}\text{C}$ with 5% CO_2 . The L929 cell line was used to evaluate the selectivity of the extracts and these assays, the anticancer drug doxorubicin was used as positive control.

Determination of cytotoxicity - MTT assay

The determination of cytotoxicity was performed by the MTT (3-(4,5-dimethyl-2-thiazolyl)-2,5-diphenyl-2H-tetrazoliumbromide) colorimetric method.²¹ The samples were tested at $100\text{ }\mu\text{g mL}^{-1}$ in six tumor cell lines for initial screening; the IC_{50} concentration was determined for those samples that showed positive results (growth inhibition $> 75\%$) in at least three cell lines. The cells were plated in 96-well plates at the following concentrations: HCT-116 / SW-620: $0.7 \times 10^5\text{ cells mL}^{-1}$; SF-295 / PC3 / L929: $0.1 \times 10^6\text{ cells mL}^{-1}$; HL-60: $3 \times 10^6\text{ cells mL}^{-1}$; RAJI: $4 \times 10^5\text{ cells mL}^{-1}$. The cells were treated with the extracts for 72 h. At the end of the treatment, the plates were centrifuged, and the supernatant removed. Then, $150\text{ }\mu\text{L}$ of MTT solution ($0.5\text{ }\mu\text{g mL}^{-1}$) was added and incubated for 3 h. After incubation, the MTT solution was removed, and the precipitated formazan was dissolved with $150\text{ }\mu\text{L}$ of dimethyl sulfoxide (DMSO). The absorbances were read using a plate spectrophotometer (Multimode Detector, DTX 880, Beckman Coulter) at 595 nm.

Statistical analysis of data activity

All experiments were performed in duplicate and repeated three times. For all samples, the selectivity index (SI) was calculated. The calculation of this index corresponds to the division between the IC_{50} value of each test compound in the L929 non-tumor cell line and the IC_{50} value of each compound in the tumor cell line

(SI = IC₅₀L929 / IC₅₀neoplastic cells).²² The experiments were analyzed according to the mean ± standard deviation (SD) of the percentage of cell growth inhibition using the GraphPad Prism software.²³

Results and Discussion

Chemical profile by UPLC-QTOF-MS^E

The ethanolic extracts of the three sections of *M. urundeuva* were obtained from the methodology described in “Sample preparation” sub-section. The extracts were analyzed by UPLC-QTOF-MS^E following the parameters described in “Chromatographic conditions” and “Mass spectrometry conditions” sub-sections only in the negative mode. In all, about 50 compounds were tentatively identified, covering the three sections of the species studied using MS and MS/MS from the chromatographic analysis (Figure 1). These results were compared to the data reported in the literature (chemotaxonomic) referring to the family (Anacardiaceae) and the genus (*Myracrodruon*) because there are few reports concerning the species. We used databases such as PubChem, ChemSpider, and Scifinder to support the results.

A wide range of phenolic compounds was identified, mainly derived from flavonoids and tannins. The predominant compounds in leaves were corilagin, firstly reported to the species, as well as geraniinic acid, and compounds well known in the literature as quercetin, gallic acids, and anacardic acids derivatives. The ethanolic extract of branches presented predominantly chlorogenic acid, quinic acid derivatives and the dimeric chalcones, urundeuvin A and

B. The bark presented mostly catechin derivatives, in addition to the compounds contained in the branch.

A fragmentation study of the possible biomarkers tentatively identified was performed, presented below. The remaining substances have been tentatively identified and are presented in Table S1 and Figure S1 (Supplementary Information section).

Hydrolysable tannins

Peaks **13**, **14** and **19** showed a precursor ion with [M – H]⁻ at *m/z* 951.0764, 633.0710 and 953.0887, respectively. Peak **13** showed fragments in 933.0729 [M – H – H₂O]⁻ and 300.9977 [M – H – 633 – H₂O]⁻ indicating losses of a water and corilagin unit with water. This compound was identified as geraniin from the literature data with a molecular mass of C₄₁H₂₈O₂₇.²⁴ Peak **14** presented MS/MS fragmentation pattern with loss of galloyl unit and water that showed fragment in 463.0564 [M – H – 152 – H₂O]⁻ and fragment in 300.9953 [M – H – 152 – 180]⁻ that corresponds to the loss of a galloyl unit with a hexose. This compound was identified as corilagin according to information previously reported in the literature,²⁴ with a molecular formula of C₂₇H₂₂O₁₈. Corilagin identification was confirmed by the comparison with molecular ion and fragmentation pattern of an analytical standard sample presented in Figure S2 (Supplementary Information section). Peak **19** showed similar fragments of the compound **13** at *m/z* 300.9953 [M – H – 3 × 152 – 180 – H₂O]⁻ and *m/z* 169.0121 as deprotonated gallic acid. This compound was identified as geraniinic acid with a molecular formula of C₄₁H₃₀O₂₇.

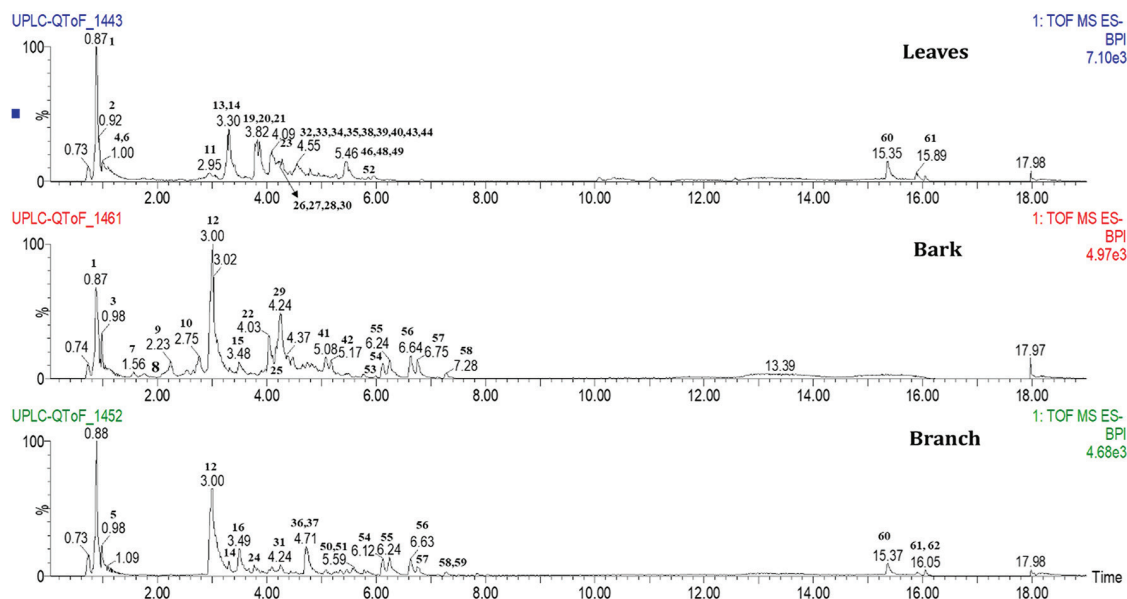


Figure 1. Typical base peak intensity (BPI) chromatograms showing the profiles of leaf, branch, and bark of *M. urundeuva* in the negative ionization mode.

The fragmentation pattern was very similar, and they are derived from the ellagitannin common in the genus *Phyllanthus*.²⁵ These compounds are being reported for the first time to the Anacardiaceae family.

Flavonols

Peaks **24**, **26**, **27**, **33** and **46** were identified as quercetin derivatives. Peak **24** presented precursor ion $[M - H]^-$ at m/z 497.1338 ($C_{22}H_{26}O_{13}$). It is an unidentified compound but may be considered quercetin derivative because it presents a characteristic fragment ion at m/z 301. Peak **26** presented precursor ion $[M - H]^-$ at m/z 463.0867 ($C_{21}H_{20}O_{12}$). This compound presented fragments in 301 $[M - H - 162]^-$ indicating loss of hexoside unit. The identification of quercetin-3-*O*-galactoside was based on the work of Erşan *et al.*²⁶ of *Pistacia vera* L., which belongs to Anacardiaceae family. Peak **27** was identified as quercetin-3-*O*-glucuronide similarly to the above compounds with precursor ion $[M - H]^-$ at m/z 477.0648 ($C_{21}H_{18}O_{13}$). The main fragment at m/z 301 $[M - H - 176]^-$ confirms the quercetin derivatives. Peak **33** presented precursor ion $[M - H]^-$ at m/z 433.0777 ($C_{20}H_{18}O_{11}$) and was identified as quercetin-3-*O*-arabinopyranoside. This determination was performed by comparison to the work of Schieber *et al.*²⁷ while was identified these compounds in the respective elution order. The main fragment at m/z 301 $[M - H - 132]^-$ confirms the quercetin derivatives by the loss of pentoside unit. Peak **46** presented the deprotonated ion $[M - H]^-$ at m/z 599.0995 ($C_{28}H_{24}O_{15}$) with fragment MS/MS at m/z 301 $[M - H - 298]^-$ referring to the loss of

one unit of raminoside-gallate, characteristic of quercitrin 2-*O*-galate, previously identified by Abu-Reidah *et al.*²⁸

Flavanols

Peaks **9** and **15** were identified as gallicocatechin derivatives. All gallicocatechin derivatives have a 125 Da fragment.²⁹ This fragmentation is shown in Figure 2 by the formation of free phenol and the non-formation of a fragment of the gallic acid, indicating the gallicocatechin and epigallicocatechin compounds. This proposal was based on Miketova *et al.*²⁹ Peak **9** presented precursor ion $[M - H]^-$ at m/z 305.0653 ($C_{15}H_{14}O_7$) which, according to the previous reference, was identified as gallicocatechin/epigallicocatechin. Identification of the correct stereochemistry is not possible only by the fragmentation pattern. The comparison with an analytical standard as well as isolation of the compound and analysis by other techniques would be necessary. Peak **15** presented precursor ion $[M - H]^-$ at m/z 457.0768. The fragmentation of peak **15** differs from **9** by the presence of the fragments at m/z 305.0664 $[M - H - 152]^-$ and 169.0125 $[M - H - 288]^-$ indicating the loss of a galloyl unit with the formation of the epigallicocatechin unit and gallic acid deprotonated unit, respectively. The formation of this compound fragments is presented in Figure 2.

Other compounds

Peak **3** presented precursor ion $[M - H]^-$ 133.0129 ($C_{14}H_8O$). In the MS/MS spectrum was at m/z 115.0042

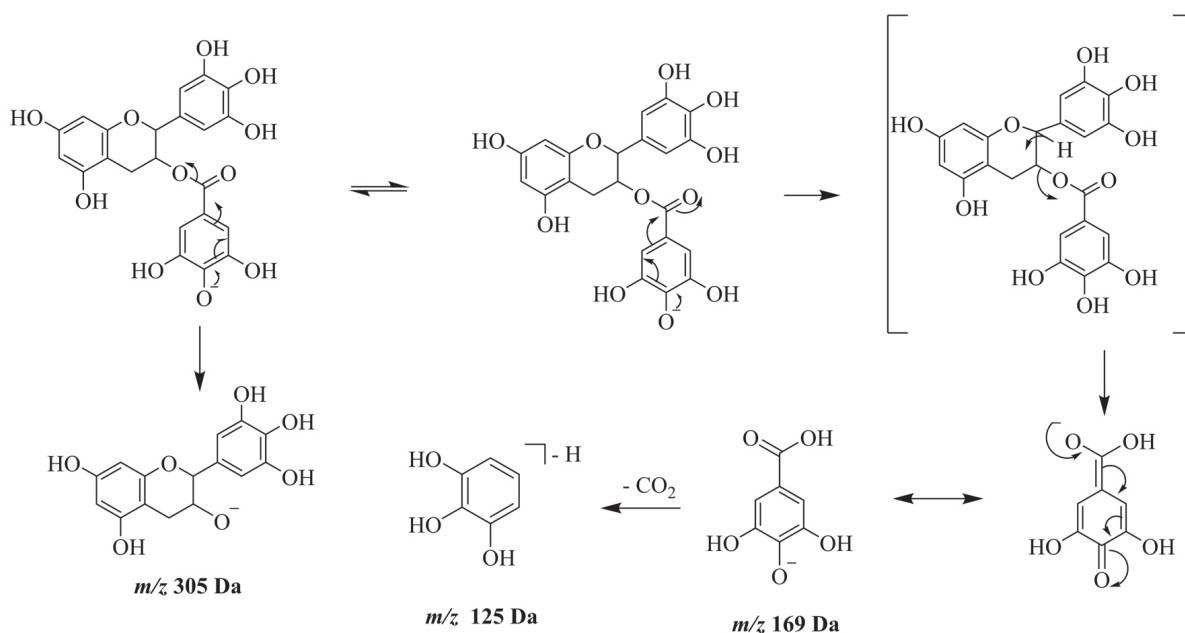


Figure 2. Proposed fragmentation of epigallicocatechin 3-*O*-gallate, collision energy ramp 20-40 eV (adapted from reference 29).

[M – H – H₂O]⁻. Based on Abu-Reidah *et al.*²⁸ work, it was identified as malic acid.²⁸

Peak **5** demonstrated in its first-order spectrum the molecular ion m/z [M – H]⁻ at 191.0192 (C₆H₈O₇). This compound presented fragment in 111.0079 Da and it was possible to identify as citric acid as suggested by Lafontaine and co-workers.³⁰

Peaks **10**, **20**, **25**, **30**, **35**, **36**, **42**, **45** and **50** presented precursor ion [M – H]⁻ at m/z 483.0813, 785.0815, 497.1295, 939.1104, 1091.1266, 341.0647, 629.1295, 603.1172 and 447.0757, respectively. All compounds showed the presence of the same fragment in 169 Da corresponding to deprotonated gallic acid. Peaks **10**, **30** and **35** presented fragments with loss of 152 Da each that correspond to galloyl unit. The precursor ions of the peaks **30** and **35** were reported in the literature to the methanolic extracts leaf of *M. urundeuva*¹ as a series of galloylglucose. Based on other reports,^{26,28} these compounds were identified as pentagalloyl hexoside and hexagalloyl hexoside, respectively.

Peak **12** showed a precursor ion [M – H]⁻ at m/z 353.0861 (C₁₆H₁₈O₉), that was identified as chlorogenic acid. To confirm the identification, the mass spectrum (MS/MS) of the analytical standard was compared with that of the extract, where similarity in retention time was observed, and the same fragment at m/z 191.0529 [M – H – 162]⁻ (Figure 3).

Peak **16** exhibited deprotonated ion [M – H]⁻ 337.0910 (C₁₆H₁₈O₈). The MS/MS spectrum presented as the base peak the ion fragment at m/z 191.0525, corresponding to the deprotonated quinic acid. The characteristic fragment ion at m/z 163.0423 corresponds, undoubtedly, to coumaric acid. According to Plazonić *et al.*,³¹ the compound was tentatively identified as 5-*p*-coumaroil quinic acid presented in Figure S3 (Supplementary Information section).

Peaks **29** and **51** showed precursor ions [M – H]⁻ at m/z 467.0902 (C₃₁H₁₆O₅) and 193.0869 (C₁₁H₁₄O₃), respectively. Despite a broad review of the literature, these compounds remained unidentified.

Peaks **55** and **56** were identified as urundevine isomers from the comparison with analytical standards. Peaks **55** and **56** presented precursor ions [M – H]⁻ at m/z 525.1268 and 525.1212 (C₃₀H₂₂O₉). Both peaks have the same fragmentation pattern corresponding to urundevine A analytical standard as can be seen in the MS/MS spectrum (Figure 4).

In addition, the assignment was corroborated with a comparison between the retention times in the extract and in the analytical standard that was 6.24 and 6.63 min; and 6.24 and 6.69 min, respectively. Therefore, these compounds were identified as urundevines A isomers II and III. Similarly, MS/MS spectra and retention time based the identification of peak **57** on the comparison of extract with urundevine B analytical standard. This peak presented precursor ion at m/z 523.1036 [M – H]⁻ and retention time of 6.78 min. Based on this information, the compounds were identified as urundevines B isomer II.

Peaks **60** and **62** were identified as anacardic acid (17:3) and (17:2),²⁶ respectively presenting at m/z 369.2400 (C₂₄H₃₄O₃) and 371.2587 (C₂₄H₃₆O₃). All of them presented a loss of one molecule of CO₂ equivalent to 44 Da in MS/MS, common in this type of molecule, giving the fragments at m/z 325.2519 and 327.2703.^{26,32}

Chemometric analysis

The main objective of using PCA analysis is to transform large amounts of complex analytical data into easily understood data.³³ The analysis allowed to observe

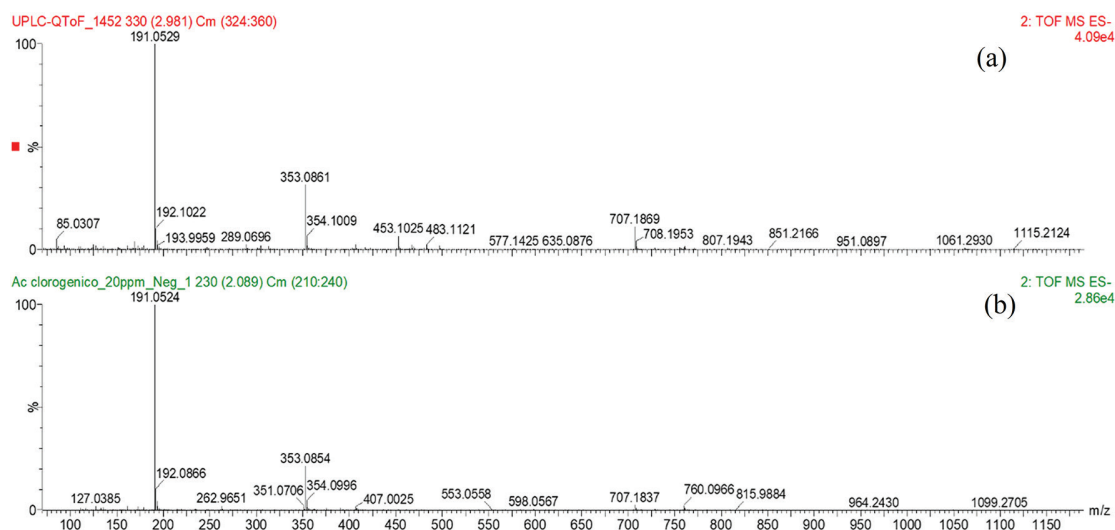


Figure 3. Spectrum in negative ion mode (ESI-MS/MS) of chlorogenic acid in the sample (a) and analytical standard (b).

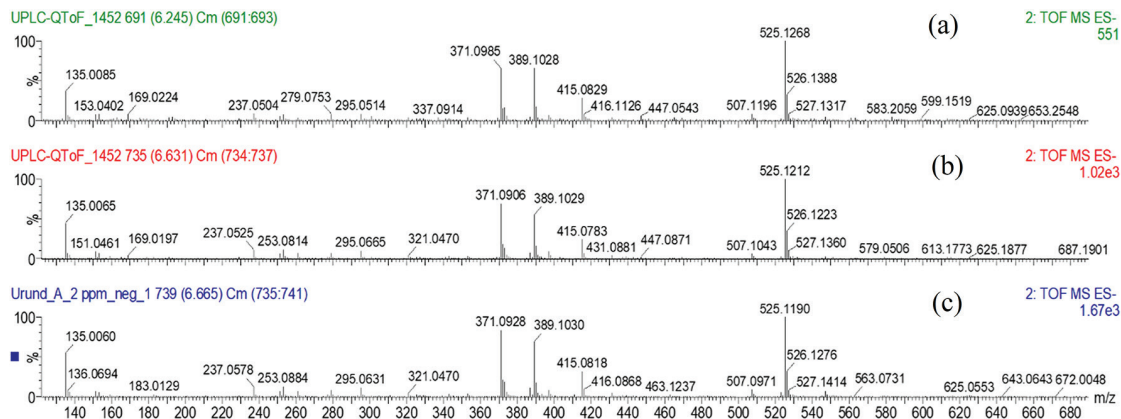


Figure 4. Spectrum in negative ion mode (ESI-MS/MS) of urundeuveine A in the samples (a) bark and (b) branch; and analytical standard (c).

more clearly the metabolic differences between the different *M. urundeuve* samples: leaf, branch, and bark. The PCA plot (Figure 5) represented 83.34% of the total variance ($R^2X [1] = 0.5959$ and $R^2X [2] = 0.2375$), using the Pareto scale. The groups formed to indicate that the secondary metabolites of the leaf, branch, and bark differ significantly. The first major component (PC1) represents the largest variation in the dataset; the branch and bark samples are on the PC1 positive side, while the leaf samples are on the negative side. The second main component (PC2) corresponds to the maximum amount of variance not explained by PC1, in which case the branch is positive for PC2 and leaf, and bark is negative. Therefore, according to the PCA data, it is evident that the three parts of the plant differ from the respective chemical profiles.

After analysis of principal components for all samples, OPLS-DA was performed among the three groups (leaf-bark, leaf-branch, and bark-branch). It was possible to verify clearly the formation of distinct groups, also observed in the PCA, demonstrating the dissimilarity between leaves, branches, and barks. In addition, in the OPLS-DA, the intra-group variation can be observed,

that is, how much the samples from the same tissues may differ from each other, and in this case, a greater homogeneity occurred in the leaf samples compared to bark and branch samples as shown in the OPLS-DA graphs (Figure 6). The good quality of the model is expressed in R^2Y (explained variance) and Q^2 (predicted variance), where the values must be above 0.5 and the closer to 1 the more reliable.³⁴

For the analysis, R^2Y and Q^2 ranged from 0.98 to 0.99, indicating that the results are highly reliable. In order to identify the metabolites that have the greatest contribution to the distinction between the parts of the plant, other statistical tools derived from the OPLS-DA were used: VIP (variable of importance in projection) and S-Plot. By employing VIP, it is possible to predict which are the most significant variables for the selection of biomarkers, in general, $VIP > 1$ is considered statistically significant.³⁵ In the present study, $VIP > 1$ and $p < 0.05$ were used. S-Plot highlights the discriminant variables, that is, those that move away from the common axis between the two groups compared. Figure 7 presents the VIP and S-Plot graphs for the leaf-bark group. The complete data containing all

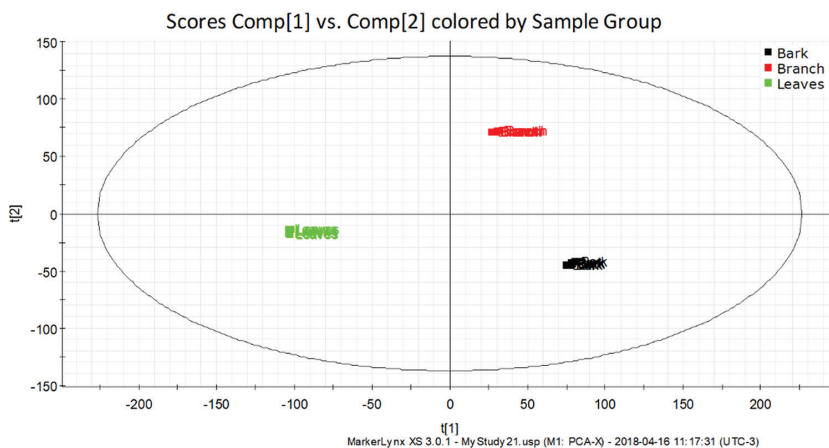


Figure 5. PCA analysis of leaf, branch and bark extracts of *M. urundeuve* (negative mode, t_R range: 0.88-17.0 min).

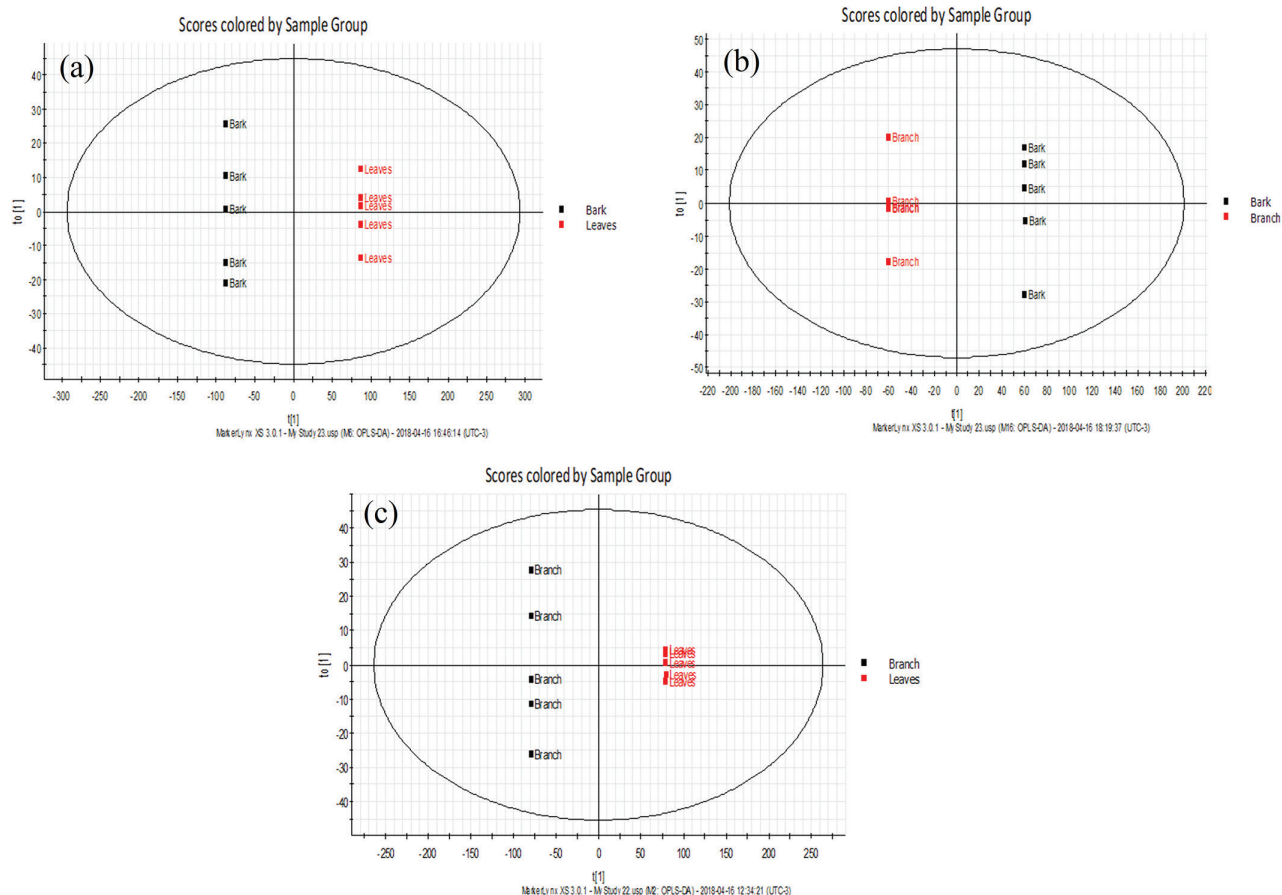


Figure 6. OPLS-DA analysis of groups: (a) leaf-branch; (b) leaf-bark; (c) bark-branch of *M. urundeuva* (negative mode, t_R range: 0.88-17.0 min).

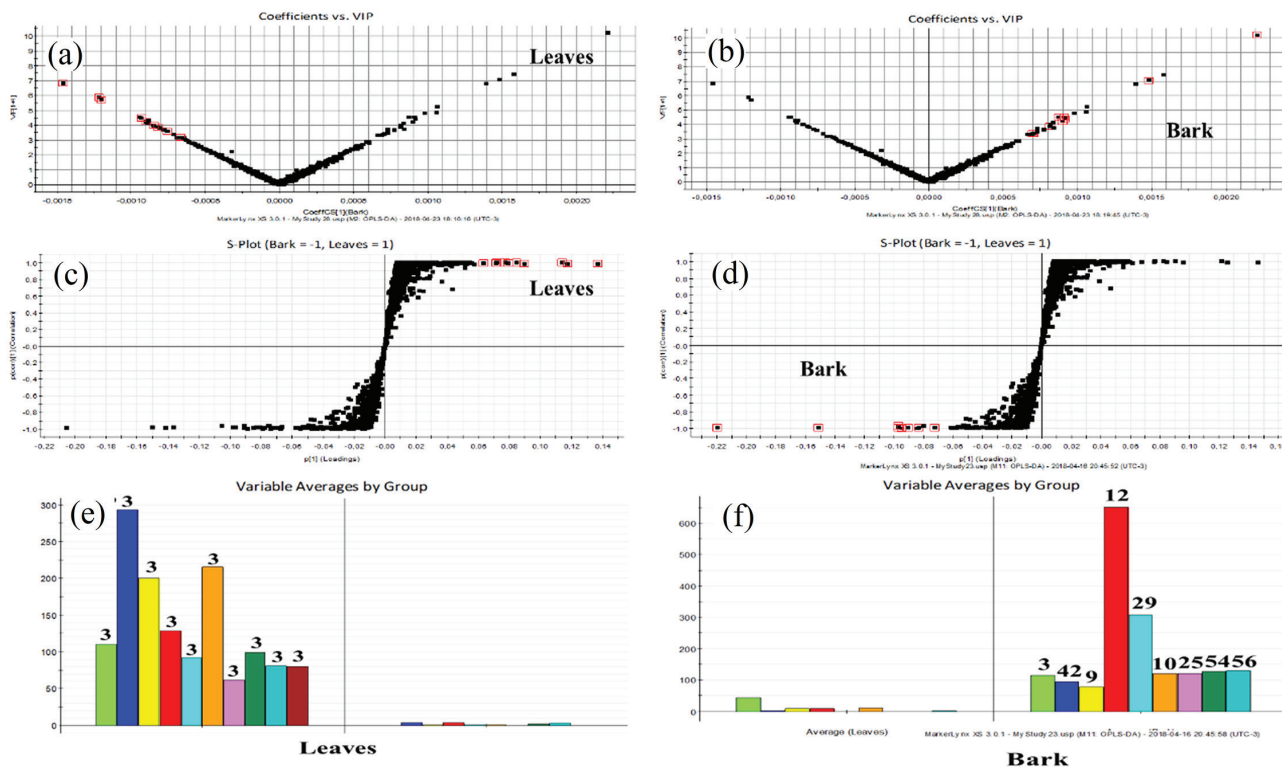


Figure 7. VIP of leaf samples (a), bark (b), S-Plot of leaf samples (c) and bark (d) and bar graph of leaf samples (e) and bark (f) of *M. urundeuva*.

other comparisons made through the S-Plot and bar charts are presented in the Figures S4 and S5 (Supplementary Information section).

After the combination of OPLS-DA, VIP and S-Plot it was possible to tentatively identify the possible biomarkers (Table 1) that may be associated with the highest cytotoxic

activity presented in all biomarkers of leaf, bark and branch extracts.

Cytotoxic activity

The screening tests (100 µg mL⁻¹) showed growth

Table 1. Biomarkers present in leaf, branch and bark ethanolic extracts of *M. urundeuva* by UPLC-QTOF-MS^E

Peak	t _R ^a / min	MS ^b [M – H] ⁻	MS/MS	Molecular formula	Tentative identification	<i>M. urundeuva</i>			VIP ^c	<i>p</i> -value ^d
						Leaves	Branch	Bark		
3	0.96	133.0129	115.0042	C ₁₄ H ₈ O	malic acid			+	3.67	1.09 × 10 ⁻⁸
5	0.98	191.0192	111.0079	C ₆ H ₈ O ₇	citric acid		+		5.00	2.55 × 10 ⁻⁵
9	2.24	305.0653	125.0232	C ₁₅ H ₁₄ O ₇	gallic acid / epigallocatechin			+	3.67	5.28 × 10 ⁻¹⁰
10	2.77	483.0813	331.0816; 169.0153	C ₂₀ H ₂₀ O ₁₄	digalloyl hexoside			+	4.59	1.34 × 10 ⁻⁹
12	3.00	353.0896	191.0493; 127.0390	C ₁₆ H ₁₈ O ₉	chlorogenic acid ^e		+	++	10.52 ^f	2.55 × 10 ^{-7f}
13	3.26	951.0764	933.0729; 300.9977	C ₄₁ H ₂₈ O ₂₇	geraniin ^g	+			3.87	2.54 × 10 ⁻⁸
14	3.30	633.0744	463.0564; 300.9948	C ₂₇ H ₂₂ O ₁₈	corilagin ^g	+			6.23	6.28 × 10 ⁻⁶
15	3.48	457.0768	305.0664; 169.0125; 125.0248	C ₂₂ H ₁₈ O ₁₁	epigallocatechin-3- <i>O</i> -gallate ^g			+	3.77	1.66 × 10 ⁻¹⁰
16	3.50	337.0910	191.0525; 163.0423; 93.0346	C ₁₆ H ₁₈ O ₈	5- <i>p</i> -courmaroil quinic acid ^g		+		5.34	6.95 × 10 ⁻¹²
19	3.80	953.0887	300.9953; 169.0121	C ₄₁ H ₃₀ O ₂₇	geraniinic acid ^g	+			5.96	3.56 × 10 ⁻⁷
20	3.87	785.0815	300.9965; 169.0135	C ₃₄ H ₂₆ O ₂₂	n.i ^b	+			6.34	1.69 × 10 ⁻¹¹
24	4.08	497.1338	301.0029	C ₂₂ H ₂₆ O ₁₃	n.i ^b		+		3.18	9.02 × 10 ⁻⁹
25	4.11	497.1295	313.0569; 217.0141; 169.0131	C ₂₂ H ₂₅ O ₁₃	n.i ^b			+	4.81	4.15 × 10 ⁻¹²
26	4.20	463.0881	301.0127	C ₂₁ H ₂₀ O ₁₂	quercetin 3- <i>O</i> -glucoside (isoquercitrin) ^g	+			3.77	2.46 × 10 ⁻¹⁰
27	4.21	477.0648	301.0137; 151.0047	C ₂₁ H ₁₈ O ₁₃	quercetin - <i>O</i> -glucuronide ^g	+			4.00	1.06 × 10 ⁻⁹
29	4.24	467.0902	357.0596; 217.0108	C ₃₁ H ₁₆ O ₅	gallic acid derivative I			+	7.70	4.34 × 10 ⁻⁹
30	4.27	939.1104	769.0933; 617.0834; 169.0125	C ₄₁ H ₃₂ O ₂₆	pentagalloyl hexoside	+			3.19	7.69 × 10 ⁻⁸
33	4.49	433.0777	301.0091; 300.0276	C ₂₀ H ₁₈ O ₁₁	quercetin 3- <i>O</i> - arabinopyranoside ^g	+			3.32	1.28 × 10 ⁻⁹
35	4.58	1091.1266	939.1158; 769.0972	C ₄₈ H ₃₆ O ₃₀	hexagalloyl hexoside	+			4.11	4.1 × 10 ⁻⁷
36	4.74	341.0647	217.0136; 189.0191; 169.0123	C ₁₈ H ₁₄ O ₇	n.i ^b		+		6.72	3.35 × 10 ⁻¹²
42	5.18	629.1295	519.0919; 467.1021; 169.0126	C ₃₃ H ₂₆ O ₁₃	n.i ^b			+	4.22	5.50 × 10 ⁻⁹
45	5.34	603.1172	341.0698; 323.0532; 169.0111	C ₃₁ H ₂₄ O ₁₃	n.i ^b			+	3.35	3.21 × 10 ⁻⁸
46	5.44	599.0995	301.0322	C ₂₈ H ₂₄ O ₁₅	quercitrin - <i>O</i> -galate isomer I ^g	+			4.79	5.02 × 10 ⁻¹¹
50	5.59	447.0757	295.0283; 169.0137	C ₂₄ H ₁₆ O ₉	n.i ^b		+		3.41	3.35 × 10 ⁻¹²
51	5.79	193.0869	178.0597; 163.0395	C ₁₁ H ₁₄ O ₃	n.i ^b		+		3.12	9.04 × 10 ⁻⁹
55	6.24	525.1195	389.1032; 371.0961; 135.0076	C ₃₀ H ₂₂ O ₉	urundeuvine A ^e		+	++	4.74 ^f	5.59 × 10 ^{-8f}
56	6.63	525.1180	389.1053; 371.0939; 135.0079	C ₃₀ H ₂₂ O ₉	urundeuvine A isomer ^e		+	++	4.79 ^f	1.18 × 10 ^{-12f}
57	6.78	523.1036	521.0576; 387.0816; 371.0969; 135.0080	C ₃₀ H ₂₀ O ₉	urundeuvine B ^e		+		3.89	1.3 × 10 ⁻⁸
60	15.38	369.2400	325.2519; 255.2318; 183.0120; 133.0716	C ₂₄ H ₃₄ O ₃	anacardic acid (17:3) ^g	+	+		1.89	0.0324
62	16.05	371.2587	327.2703	C ₂₄ H ₃₆ O ₃	anacardic acid (17:2) ^g	+	+		4.10	1.94 × 10 ⁻⁵

^aRetention time; ^bmass spectrometry; ^cvariable of importance in projection; ^dprobability value; ^ecompounds that were compared with an analytical standard; ^fVIP and *p*-value values of the part of the plant that has the highest relative concentration of the compound tentatively identified; ^gcompounds reported for the first time, in *Myracrodruon urundeuva*; ^hnot identified. +: lower relative concentration; ++: highest relative concentration.

inhibition above 70% in all cell lines exposed to ethanolic bark extracts. The leaf extracts were toxic only to leukemia (HL-60) cells. Table 2 shows the results of the cytotoxicity assays.

After initial screening, IC₅₀ tests were performed with the bark and leaf extracts. The extracts showed higher cytotoxic potential against the leukemic cell line with IC₅₀ ranging from 17.46 (bark) to 18.55 µg mL⁻¹ (leaf) (Table 3).

Other studies show *in vitro* cytotoxic effects of plant ethanolic extracts. IC₅₀ of 38.1 µg mL⁻¹ was found after treatment of leukemic cells with ethanolic extract of *M. urundeuva* seeds.⁷ The authors showed DNA fragmentation and mitochondrial depolarization caused by seed extracts.

Studies on the inhibition of tumor cells growth under the effect of the ethanolic extract of *M. urundeuva* were carried out.⁵ IC₅₀ values for this study were 7.4 µg mL⁻¹ against HL-60 and 8.5 µg mL⁻¹ against SW-1573. Bark ethanolic extract, for example, presented 95.6% growth inhibition for the breast, colon, and glioblastoma lines.⁶

Viana *et al.*⁴ demonstrated that hydroethanolic extract of *M. urundeuva* bark exerts anti-inflammatory and analgesic effects related to chalcones. The extracts have antioxidant properties attributed to flavonoids.³⁶ Souza *et al.*⁸ demonstrated the anti-inflammatory and protective effects against gastric ulcer in mice or rat after treatment with fraction rich in tannins extracted from the "aroeira" using ethyl acetate as the solvent.

The selectivity index (SI) of each sample was evaluated.

The SI measures how much a compound is active against tumor cells without causing damage to non-tumor ones, and it is interesting when it presents values greater than 2.0.³⁷ In the cell lines tested the leaf extract showed selectivity to the HL-60 line with an index higher than 2. This result can be correlated by the possible biomarkers, tentatively identified via chemometric analysis, present in the leaf, such as corilagin, geraniin, geraniic acid, quercetin derivatives, among others.

Corilagin, a compound well described in the literature, presents a variety of pharmacological effects, such as anti-tumor,³⁸ anti-inflammatory,³⁹ antioxidant,⁴⁰ and hepatoprotective.⁴¹ Also, the literature reports good antitumor activity along with low toxicity to healthy cells and tissues, making corilagin a promising anticancer lead molecule.⁴² The geraniin, another possible adjuvant compound, is known to exert antitumor,⁴³ antibacterial,⁴⁴ antioxidant⁴⁵ and antiviral activities.^{46,47} It has already been reported in the literature⁴⁸⁻⁵⁰ that quercetin and its derivatives are well known for their antioxidant, antihistaminic and anti-inflammatory properties. The quercetin is being considered a promising new chemotherapeutic agent, and several studies are underway to explore molecules derived from quercetin for cancer-directed chemotherapy.⁵¹ In addition, there may be other substances, as well as their synergistic compounds, that play significant roles in the reported bioactivity but were not identified in the present work due to limitations of the technique chosen.

The bark showed very promising results. The SI was

Table 2. Average cell growth inhibition (GI) of *M. urundeuva* extracts at 100 µg mL⁻¹

Extract	HL-60 ^a / %	SD ^b / %	HCT-116 ^c / %	SD ^b / %	SF-295 ^d / %	SD ^b / %	RAJI ^a / %	SD ^b / %
Leaves	86.14	4.62	69.48	4.23	64.93	3.24	41.84	3.38
Branch	41.98	4.94	–	–	10.52	5.47	–	–
Bark	88.56	1.99	80.47	2.46	83.41	1.87	77.56	0.70

^aLeukemia; ^bstandard deviation; ^ccolorectal; ^dglioblastoma. Results are expressed as mean percent cell growth inhibition (IC) and standard deviation for two independent experiments in triplicate.

Table 3. IC₅₀ values with a 95% confidence interval of *M. urundeuva* extracts in tumor and non-tumoral cell lines

Sample	IC ₅₀ ^a (interval) / (µg mL ⁻¹)						
	SF-295 (glioblastoma)	PC3 (prostate)	HL-60 (leukemia)	RAJI (leukemia)	HCT-116 (colorectal)	SW-620 (colorectal)	L929 (murine fibroblast)
Bark	38.86 (31.03-48.67)	> 100	17.46 (16.29-18.70)	65.73 (56.77-76.11)	32.50 (29.44-35.87)	40.68 (30.64-54.00)	> 100
Leaf	72.61 (55.80-94.49)	> 100	18.55 (14.80-23.26)	55.42 (44.49-69.04)	55.60 (47.24-65.45)	> 100	55.42 (47.06-65.26)
Doxorubicin ^b	0.25 (0.22-0.28)	0.44 (0.34-0.54)	0.01 (0.005-0.01)	0.46 (0.45-0.47)	0.11 (0.08-0.14)	0.03 (0.02-0.05)	0.99 (0.92-1.08)

^aHalf maximal inhibitory concentration (IC₅₀) values with a 95% confidence interval obtained by non-linear regression from three independent experiments performed in duplicate on six tumor lines and one non-tumor line; ^bdoxorubicin was used as a positive control.

higher than 2 for all cell lines tested except for prostate (PC3) and leukemia (RAJI) (Table S2, Supplementary Information section). Eventually, in the extract, it is possible to highlight the discriminant biomarkers belonging to bark that may be related to the cytotoxic activity presented, such as urundevine A, galloy derivatives, catechins, and phenolic acids. Bandeira *et al.*⁵² had successfully isolated dimeric chalcones: urundevine A, B, C and matosin from the internal bark of *M. urundeuva*, and Souza *et al.*⁸ had demonstrated antimicrobial and anti-inflammatory activity for *M. urundeuva*. The literature also has evidence for the antitumor activity being promoted by specific classes of flavonoids such as chalcones, flavonones, and flavones. Many derivatives of these classes showed significant activities against some tumoral cell lines such as human colon, breast, and kidneys.⁵³ Different polyphenols from “aroeira-vermelha” (*Schinus terebinthifolius* Raddi) induced cell death of human prostate carcinoma and were considered capable of modulating cell proliferation according to the test concentration.⁵³ The use of catechins has shown inhibition of prostate and colon cancer.^{54,55} The combination of classical chemotherapy with nutrients and especially with polyphenols may decrease the pressure and the adverse effects of the antineoplastic drug.⁵⁶ Therefore, three of the polyphenols present in the bark of the “aroeira” tree are promising compounds for isolation or synthesis into the development of phytopharmaceutical products from natural extracts. Chemical investigations of this extract can be a promising strategy for the discovery of phytotherapeutic agents. Also, the chemical profile comparison of bark and branch extracts revealed compounds that may be important for their biological activity.

Conclusions

From a simple and rapid extraction method, it was possible to trace the chemical profile of the three plant organs of *M. urundeuva* as leaf, branch and bark using the analytical technique UPLC-QTOF-MS^E, which allowed the tentative identification of 50 compounds which covered several classes of compounds as flavonoids, flavanoids, hydrolysable tannins and anacardic acid. From the multivariate data analyses presented, it was possible to have information about the metabolic differences between the extracts compared. Such an association has been significant in the discussion of observed activities because the extracts obtained different responses against the tested lines.

The bark and leaf extract showed high toxicity and low IC₅₀ values against the HL-60 (leukemia), HCT-116 (human colon) and RAJI (leukemia) cell lines compared to the branch. The higher relative concentration of compounds

derived from quercetin, galloy derivatives, and phenolic acids present in these extracts may contribute to the understanding of the observed high cytotoxic activity. Some of the compounds identified, such as quercetin derivatives, corilagin, and chlorogenic acid, already has activities recognized as anti-tumor, antioxidants, and anti-inflammatory, among others, which may explain the promising activities observed here compared to literature.

Besides, through the statistical analysis, it was possible to observe the separation of the groups concerning each part of the plant and the identification of the 30 possible biomarkers. Therefore, this metabolic study notes the importance and value of the *M. urundeuva* plant as a possible source of secondary metabolites that are likely to act to inhibit certain types of cancer cells.

Supplementary Information

Table of tentative identification of the secondary metabolites present in the ethanolic extracts of leaf, bark, and branches of *M. urundeuva*; values of IC₅₀ (non-tumor cells) / IC₅₀ (tumor cell) selectivity index in tumor lines; spectrum in negative ion mode (ESI-MS/MS) of corilagin and 5-*p*-courmaroil quinic acid; S-Plot of leaf samples, branch, bar graph of leaf samples and branch; S-Plot of bark samples, branch and bar graph of bark samples and branch of *M. urundeuva* are available free of charge at <http://jbcbs.sbgq.org.br> as PDF file.

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